

**ACADEMIC REGULATIONS  
COURSE STRUCTURE  
AND  
DETAILED SYLLABUS**

**CHEMICAL ENGINEERING**

*For*

**B.TECH. FOUR YEAR DEGREE COURSE**  
*(Applicable for the batches admitted from 2015-2016)*



**JNTUH COLLEGE OF ENGINEERING HYDERABAD**  
**(Autonomous)**  
Kukatpally, Hyderabad - 500085  
TELANGANA, INDIA



**JNTUH COLLEGE OF ENGINEERING HYDERABAD  
(Autonomous)  
Kukatpally, Hyderabad-500 085**

**ACADEMIC REGULATIONS 2015  
for CBCS Based B.Tech. PROGRAMMES**

(Effective for the students admitted into I year from the Academic Year **2015-16** and onwards)

**1.0 Under-Graduate Degree Programme in Engineering & Technology (UGP in E&T)**

JNTUH offers 4 Year (8 Semesters) **Bachelor of Technology** (B.Tech.) Degree Programme, under Choice Based Credit System (CBCS) at its Constituent Autonomous College - JNTUH College of Engineering, Hyderabad, with effect from the Academic Year 2015 - 16 onwards, in the following Branches of Engineering ...

<b>S.No.</b>	<b>Branch</b>
I.	Civil Engineering
II.	Computer Science and Engineering
III.	Electrical and Electronics Engineering
IV.	Electronics and Communication Engineering
V.	Mechanical Engineering
VI.	Metallurgical Engineering
VII.	Chemical Engineering

**2.0 Eligibility for Admission**

**2.1** Admission to the UGP shall be made either on the basis of the merit rank obtained by the qualifying candidate at an Entrance Test conducted by the Telangana State Government (EAMCET), OR the University, OR on the basis of any other order of merit approved by the University, subject to reservations as prescribed by the Government from time to time.

**2.2** The medium of instructions for the entire UGP in E&T will be ENGLISH only.

**3.0 B.Tech. Programme (UGP) Structure**

**3.1** The B.Tech. Programmes of JNTUH-CEH are of Semester Pattern, with 8 Semesters constituting 4 Academic Years, each Academic Year having TWO Semesters (First/Odd and Second/Even Semesters). Each Semester shall be of 22 Weeks duration (inclusive of Examinations), with a minimum of 90 Instructional Days per Semester.

**3.2** UGC/ AICTE specified Definitions/ Descriptions are adopted appropriately for various terms and abbreviations used in these Academic Regulations/ Norms, which are as listed below.

**3.2.1 Semester Scheme:**

Each UGP is of 4 Academic Years (8 Semesters), with the year being divided into two Semesters of 22 weeks ( $\geq 90$  working days) each, each Semester having - 'Continuous Internal Evaluation (CIE)' and 'Semester End Examination (SEE)'. Choice Based Credit System (CBCS) and Credit Based Semester System (CBSS) as denoted by UGC, and Curriculum/ Course Structure as suggested by AICTE are followed.

**3.2.2 Credit Courses:**

All Subjects/ Courses are to be registered by a student in a Semester to earn Credits. Credits shall be assigned to each Subject/ Course in a L: T: P: C (Lecture Periods: Tutorial Periods: Practicals Periods : Credits) Structure, based on the following general pattern ..

- One Credit - for One hour/ Week/ Semester for Theory/ Lecture (L) Courses; and,
- One Credit - for Two hours/ Week/ Semester for Laboratory/ Practical (P) Courses or Tutorials (T). Other student activities like NCC, NSS, NSO, Study Tour, Guest Lecture etc., and identified Mandatory Courses will not carry Credits.

**3.2.3 Subject/ Course Classification:**

All Subjects/ Courses offered for the UGP are broadly classified as : (a) Foundation Courses (FnC), (b) Core Courses (CoC), and (c) Elective Courses (EłC).

- Foundation Courses (FnC) are further categorized as : (i) HS (Humanities and Social Sciences), (ii) BS (Basic Sciences), and (iii) ES (Engineering Sciences);
- Core Courses (CoC) and Elective Courses (EłC) are categorized as PS (Professional Subjects), which are further subdivided as – (i) PC (Professional/ Departmental Core) Subjects, (ii) PE (Professional/ Departmental Electives), (iii) OE (Open Electives); and (iv) Project Works (PW);
- Minor Courses (1 or 2 Credit Courses, belonging to HS/ BS/ ES/ PC as per relevance); and
- Mandatory Courses (MC - non-credit oriented).

**3.2.4 Course Nomenclature:**

The Curriculum Nomenclature or Course-Structure Grouping for the each of the UGP E&T (B.Tech. Degree Programmes), is as listed below (along with AICTE specified % Range of Total Credits)...

S. No.	Broad Course Classification	Course Group/ Category	Course Description	Range of Credits
1)	Foundation Courses (FnC)	BS – Basic Sciences	Includes - Mathematics, Physics and Chemistry Subjects	15% - 20%
2)		ES - Engineering Sciences	Includes fundamental engineering subjects	15% - 20%
3)		HS – Humanities and Social Sciences	Includes subjects related to Humanities, Social Sciences and Management	5% - 10%
4)	Core Courses (CoC)	PC – Professional Core	Includes core subjects related to the Parent Discipline/ Department/ Branch of Engg.	30% - 40%

5)	Elective Courses (EłC)	PE – Professional Electives	Includes Elective subjects related to the Parent Discipline/ Department/ Branch of Engg.	10% - 15%
6)		OE – Open Electives	Elective subjects which include inter-disciplinary subjects or subjects in an area outside the Parent Discipline/ Department/ Branch of Engg.	5% - 10%
7)	Core Courses	Project Work	B.Tech. Project or UG Project or UG Major Project	10% - 15%
8)		Industrial Training/ Mini-Project	Industrial Training/ Internship/ UG Mini-Project/ Mini-Project	
9)		Seminar	Seminar/ Colloquium based on core contents related to Parent Discipline/ Department/ Branch of Engg.	
10)		Minor Courses	1 or 2 Credit Courses (subset of HS)	included
11)		Mandatory Courses (MC)	Mandatory Courses (non-credit)	-
Total Credits for UGP (B. Tech.) Programme				192 (100%)

**4.0 Course Work**

- 4.1** A student, after securing admission, shall pursue the B.Tech. UGP in a minimum period of 4 Academic Years, and a maximum period of 8 Academic Years (starting from the Date of Commencement of I Year).
- 4.2** Each student shall Register for and Secure the specified number of Credits required for the completion of the UGP and Award of the B.Tech. Degree in respective Branch of Engineering.
- 4.3** Each Semester is structured to provide typically 24 Credits (24 C), totaling to 192 Credits (192 C) for the entire B.Tech. Programme.

**5.0 Course Registration**

- 5.1** A 'Faculty Advisor or Counselor' shall be assigned to each student, who will advise him about the UGP, its Course Structure and Curriculum, Choice/Option for Subjects/ Courses, based on his competence, progress, pre-requisites and interest.
- 5.2** Academic Section of the College invites 'Registration Forms' from students apriori (before the beginning of the Semester), through 'ON-LINE SUBMISSIONS', ensuring 'DATE and TIME Stamping'. The ON-LINE Registration Requests for any 'CURRENT SEMESTER' shall be completed BEFORE the commencement of SEEs (Semester End Examinations) of the 'PRECEDING SEMESTER'.
- 5.3** A Student can apply for ON-LINE Registration, ONLY AFTER obtaining the 'WRITTEN APPROVAL' from his Faculty Advisor, which should be submitted to the College Academic Section through the Head of Department (a copy of the same being retained with Head of Department, Faculty Advisor and the Student).
- 5.4** A Student may be permitted to Register for his Subjects/ Course of CHOICE with a typical total of 24 Credits per Semester (Minimum being 20 C and Maximum being 28 C, permitted deviation being  $\pm 17\%$ ), based on his PROGRESS and SGPA/ CGPA, and completion of the 'PRE-REQUISITES' as indicated for various Subjects/ Courses, in the Department Course Structure and Syllabus contents. However, a MINIMUM of 20 Credits per Semester must be registered to ensure the 'STUDENTSHIP' in any Semester.
- 5.5** Choice for 'additional Subjects/ Courses' to reach the Maximum Permissible Limit of 28 Credits (above the typical 24 Credit norm) must be clearly indicated, which needs the specific approval and signature of the Faculty Advisor/ Counselor.
- 5.6** If the Student submits ambiguous choices or multiple options or erroneous entries - during ON-LINE Registration for the Subject(s) / Course(s) under a given/ specified Course Group/ Category as listed in the Course Structure, only the first mentioned Subject/ Course in that Category will be taken into consideration.

- 5.7** Subject/ Course Options exercised through ON-LINE Registration are final and CAN NOT be changed, and CAN NOT be inter-changed; further, alternate choices will also not be considered. However, if the Subject/ Course that has already been listed for Registration (by the Head of Department) in a Semester could not be offered due to any unforeseen or unexpected reasons, then the Student shall be allowed to have alternate choice - either for a new Subject (subject to offering of such a Subject), or for another existing Subject (subject to availability of seats), which may be considered. Such alternate arrangements will be made by the Head of Department, with due notification and time-framed schedule, within the FIRST WEEK from the commencement of Class-work for that Semester.
- 5.8** Dropping of Subjects/ Courses may be permitted, ONLY AFTER obtaining prior approval from the Faculty Advisor (subject to retaining a minimum of 20 C), 'within 15 Days of Time' from the beginning of the current Semester.
- 5.9** For Mandatory Courses like NCC/ NSS/ NSO etc., a 'Satisfactory Participation Certificate' from the concerned authorities for the relevant Semester is essential. No Marks or Grades or Credits shall be awarded for these activities.
- 6.0 Subjects/ Courses to be offered**
- 6.1** A typical Section (or Class) Strength for each Semester shall be 60.
- 6.2** A Subject/ Course may be offered to the Students, ONLY IF a Minimum of 20 Students ( $1/3$  of the Section Strength) opt for the same. The Maximum Strength of a Section is limited to 80 ( $60 + 1/3$  of the Section Strength).
- 6.3** More than ONE TEACHER may offer the SAME SUBJECT (Lab./ Practicals may be included with the corresponding Theory Subject in the same Semester) in any Semester. However, selection choice for students will be based on - 'FIRST COME FIRST SERVE Basis and CGPA Criterion' (ie., the first focus shall be on early ON-LINE ENTRY from the student for Registration in that Semester, and the second focus, if needed, will be on CGPA of the student).

**6.4** If more entries for Registration of a Subject come into picture, then the concerned Head of Department shall take necessary action, whether to offer such a Subject/ Course for TWO (or multiple) SECTIONS or NOT .

**6.5** In case of options coming from Students of other Departments/ Branches/ Disciplines (not considering OPEN ELECTIVES), PRIORITY shall be given to the student of the 'Parent Department' first.

## **7.0 Attendance Requirements**

**7.1** A student shall be eligible to appear for the End Semester Examinations, if he acquires a minimum of 75% of attendance in aggregate of all the Subjects/ Courses (excluding Mandatory or Non-Credit Courses) for that Semester.

**7.2** Condoning of shortage of attendance in aggregate up to 10% (65% and above, and below 75%) in each Semester may be granted by the College Academic Committee on genuine and valid grounds, based on the student's representation with supporting evidence.

**7.3** A stipulated fee shall be payable towards condoning of shortage of attendance.

**7.4** Shortage of Attendance below 65% in aggregate shall in NO case be condoned.

**7.5** Students, whose shortage of attendance is not condoned in any Semester, are not eligible to take their End Examinations of that Semester, they get detained and their registration for that Semester shall stand cancelled. They will not be promoted to the next Semester. They may seek re-registration for all those Subjects registered in that Semester in which he got detained, by seeking re-admission for that Semester as and when offered; in case if there are any Professional Electives and/ or Open Electives, the same may also be re-registered if offered, however, if those Electives are not offered in later Semesters, then alternate Electives may be chosen from the SAME set of Elective Subjects offered under that category.

## **8.0 Academic Requirements**

The following Academic Requirements have to be satisfied, in addition to the Attendance Requirements mentioned in Item No.7.

**8.1** A student shall be deemed to have satisfied the Academic Requirements and earned the Credits allotted to each Subject/ Course, if he secures not less than 35% marks (25 out of 70 marks) in the End Semester Examination, and a minimum of 40% of marks in the sum total of the CIE (Continuous Internal Evaluation) and SEE (Semester End Examination) taken together; in terms of Letter Grades, this implies securing P Grade or above in that Subject/ Course.

**8.2** A student shall be deemed to have satisfied the Academic Requirements and earned the Credits allotted to - Industry oriented Mini-Project/ Seminar, if he secures not less than 40% of the total marks (40 marks) to be awarded for each. The student would be treated as failed, if he - (i) does not submit a report on his Industry oriented Mini-Project, or does not make a presentation of the same before the Evaluation Committee as per schedule, or (ii) does not present the Seminar as required in the IV year II Semester, or (iii) secures less than 40% of marks (40 marks) in Industry oriented Mini-Project/ Seminar evaluations. He may reappear once for each of the above evaluations, when they are scheduled again; if he fails in such 'one reappearance' evaluation also, he has to reappear for the same in the next subsequent Semester, as and when it is scheduled.

**8.3** A Student will not be promoted from I Year to II Year, unless he fulfils the Attendance and Academic Requirements and secures a total of 24 Credits out of 48 Credits of I Year, from all the relevant regular and supplementary examinations, whether he takes those examinations or not.

**8.4** A Student will not be promoted from II Year to III Year, unless he fulfils the Attendance and Academic Requirements and secures a total of 43 Credits out of 72 Credits upto II Year I Semester, from all the relevant regular and supplementary examinations, whether he takes those examinations or not.

- 8.5** A Student will not be promoted from III Year to IV Year, unless he fulfils the Attendance and Academic Requirements and secures a total of 72 Credits out of 120 Credits upto III Year I Semester, from all the relevant regular and supplementary examinations, whether he takes those examinations or not.
- 8.6** A Student shall - register for all Subjects covering 192 Credits as specified and listed (with the relevant Course/ Subject Classifications as mentioned) in the Course Structure, put up all the Attendance and Academic requirements for 192 Credits securing a minimum of P Grade (Pass Grade) or above in each Subject, and 'earn ALL 192 Credits securing SGPA  $\geq 5.0$  (in each Semester), and CGPA (at the end of each successive Semester)  $\geq 5.0$ ', to successfully complete the UGP.
- 8.7** After securing the necessary 192 Credits as specified for the successful completion of the entire UGP, an exemption of 8 secured Credits (in terms of two of their corresponding Subjects/Courses) may be permitted for optional drop out from these 192 Credits earned; resulting in 184 Credits for UGP performance evaluation, i.e., the performance of the Student in these 184 Credits shall alone be taken into account for the calculation of 'the final CGPA (at the end of UGP, which takes the SGPA of the IV Year II Semester into account)', and shall be indicated in the Grade Card of IV Year II Semester; however, the Student's Performances in the earlier individual Semesters, with the corresponding SGPA and CGPA for which already Grade Cards are given, will not be altered. Further, optional drop out for such 8 secured Credits shall not be allowed for Subjects/ Courses listed as ... i) Laboratories/ Practicals, Industrial Training/ Mini-Project, iii) Seminar, iv) Major Project.
- 8.8** If a Student registers for some more 'extra Subjects' (in the parent Department or other Departments/Branches of Engg.) other than those listed Subjects totaling to 192 Credits as specified in the Course Structure of his Department, the performances in those 'extra Subjects' (although evaluated and graded using the same procedure as that of the required 192 Credits) will not be taken into account while calculating the SGPA and CGPA. For such 'extra Subjects'

registered, % marks and Letter Grade alone will be indicated in the Grade Card, as a performance measure, subject to completion of the Attendance and Academic Requirements as stated in Items 7 and 8.1 – 8.7 above.

- 8.9** Students who fail to earn 192 Credits as per the Course Structure, and as indicated above, within 8 Academic Years from the Date of Commencement of their I Year shall forfeit their seats in B.Tech. Programme and their admissions shall stand cancelled.
- 8.10** When a Student is detained due to shortage of attendance in any Semester, he may be re-admitted into that Semester, as and when offered, with the Academic Regulations of the Batch into which he gets readmitted. However, no Grade Allotments or SGPA/ CGPA calculations will be done for that entire Semester in which he got detained.
- 8.11** When a Student is detained due to lack of Credits in any year, he may be readmitted in the next year, after fulfilment of the Academic Requirements, with the Academic Regulations of the Batch into which he gets readmitted.
- 8.12** A student eligible to appear in the End Semester Examination in any Subject/ Course, but absent at it or failed (thereby failing to secure P Grade or above), may reappear for that Subject/ Course at the supplementary examination (SEE) as and when conducted. In such cases, his Internal Marks (CIE) assessed earlier for that Subject/ Course will be carried over, and added to the Marks to be obtained in the SEE supplementary examination, for evaluating his performance in that Subject.
- 9.0 Evaluation - Distribution and Weightage of Marks**
- 9.1** The performance of a student in each Semester shall be evaluated Subject-wise (irrespective of Credits assigned) with a maximum of 100 marks for Theory or Practicals or Seminar or Drawing/Design or Industry oriented Mini-Project or Minor Course, etc; however, the B.Tech. Project Work (Major Project) will be evaluated for 200 Marks. These evaluations shall be based on 30% CIE (Continuous Internal Evaluation) and 70% SEE (Semester End Examination), and a Letter Grade corresponding to the % marks obtained shall be given.

- 9.2** For all Subjects/ Courses as mentioned above, the distribution shall be 30 marks for CIE, and 70 marks for the SEE.
- 9.3 a)** For Theory Subjects (inclusive of Minor Courses), during the Semester, there shall be 2 mid-term examinations for 25 marks each. Each mid-term examination consists of one objective paper for 10 marks, plus one subjective paper for 15 marks, with a duration of 120 minutes (20 minutes for objective and 100 minutes for subjective papers). Further, there will be an allocation of 5 marks for Assignment. Objective paper may be set with multiple choice questions, True/ False, fill-in the blanks, matching type questions, etc. Subjective paper shall contain 5 questions, out of which the Student has to answer 3 questions, each for 5 marks.
- b)** The first mid-term examination shall be conducted for the first 50% of the syllabus, and the second mid-term examination shall be conducted for the remaining 50% of the syllabus.
- c)** First Assignment should be submitted before the conduct of the first mid-term examinations, and the Second Assignment should be submitted before the conduct of the second mid-term examinations. The Assignments shall be as specified by the concerned subject teacher.
- d)** The first mid-term examination Marks and first Assignment Marks shall make one set of CIE Marks, and the second mid-term examination Marks and second Assignment Marks shall make second set of CIE Marks; and the better of these two sets of marks shall be taken as the final marks secured by the Student towards Continuous Internal Evaluation in that Theory Subject.
- 9.4** For Practical Subjects, there shall be a Continuous Internal Evaluation (CIE) during the Semester for 30 internal marks, and 70 marks are assigned for Lab./Practical End Semester Examination (SEE). Out of the 30 marks for internals, day-to-day work in the laboratory shall be evaluated for 20 marks; and for the remaining 10 marks - two internal practical tests (each of 10 marks) shall be conducted by the concerned laboratory teacher and the better of these two tests is taken into account. The SEE for

Practicals shall be conducted at the end of the Semester by Two Examiners appointed by Head of the Department.

- 9.5** For the Subjects having Design and/or Drawing, (such as Engineering Graphics, Engineering Drawing, Machine Drawing, Production Drawing Practice, and Estimation), the distribution shall be 30 marks for CIE (20 marks for day-to-day work, and 10 marks for internal tests) and 70 marks for SEE. There shall be two internal tests in a Semester and the better of the two shall be considered for the award of marks for internal tests.
- 9.6** Open Electives: Students are to choose One Open Elective (OE-I) during III Year I Semester, one (OE-II) during III Year II Semester, and one (OE-III) in IV Year II Semester, from the list of Open Electives given. However, Students can not opt for an Open Elective Subject offered by their own (parent) Department, if it is already listed under any category of the Subjects offered by parent Department in any Semester.
- 9.7 a)** There shall be an Industry oriented Mini-Project, in collaboration with an Industry of the relevant specialization, to be registered immediately after III Year II Semester examinations, and taken up during the summer vacation for about eight weeks duration.
- b)** The Industry oriented Mini-Project shall be submitted in a Report form, and a presentation of the same shall be made before a Committee, which evaluates it for 100 marks. The Committee shall consist of Head of the Department, the supervisor of Mini-Project, and a Senior Faculty Member of the Department. There shall be no internal marks for Industry oriented Mini-Project. The Mini-Project shall be evaluated in the IV Year I Semester.
- 9.8** There shall be a Seminar Presentation in IV Year II Semester. For the Seminar, the student shall collect the information on a specialized topic, prepare a Technical Report and submit to the Department at the time of Seminar Presentation. The Seminar Presentation (along with the Technical Report) shall be evaluated by Two Faculty Members assigned by Head of the Department, for 100 marks. There shall be no SEE or external examination for Seminar.

**9.9** Each Student shall start the Project Work during the IV Year I Semester, as per the instructions of the Project Guide/ Project Supervisor assigned by the Head of Department. Out of a total 200 marks allotted for the Project Work, 60 marks shall be for CIE (Continuous Internal Evaluation and 140 marks for the SEE (End Semester Viva-voce Examination). The Project Viva-voce shall be conducted by a Committee comprising of an External Examiner, Head of the Department and Project Supervisor. Out of 60 marks allocated for CIE, 30 marks shall be awarded by the Project Supervisor (based on the continuous evaluation of student's performance throughout the Project Work period), and the other 30 marks shall be awarded by a Departmental Committee consisting of Head of the Department and Project Supervisor, based on the work carried out and the presentation made by the Student at the time of Viva-voce Examination.

**9.10** For NCC/ NSS/ NSO types of Courses, and/or any other Mandatory Non-Credit Course offered in a Semester, a 'Satisfactory Participation Certificate' shall be issued to the Student from the concerned authorities, only after securing  $\geq 65\%$  attendance in such a Course. No marks or Letter Grade shall be allotted for these activities.

## **10.0 Grading Procedure**

**10.1** Marks will be awarded to indicate the performance of each student in each Theory Subject, or Lab/Practicals, or Seminar, or Project, or Mini-Project, Minor Course etc., based on the % marks obtained in CIE + SEE (Continuous Internal Evaluation + Semester End Examination, both taken together) as specified in Item 9 above, and a corresponding Letter Grade shall be given.

**10.2** As a measure of the student's performance, a 10-point Absolute Grading System using the following Letter Grades (UGC Guidelines) and corresponding percentage of marks shall be followed ...

<b>% of Marks Secured (Class Intervals)</b>	<b>Letter Grade (UGC Guidelines)</b>	<b>Grade Points</b>
<b>80% and above (<math>\geq 80\%</math>, <math>\leq 100\%</math>)</b>	<b>O (Outstanding)</b>	<b>10</b>
<b>Below 80% but not less than 70% (<math>\geq 70\%</math>, <math>&lt; 80\%</math>)</b>	<b>A<sup>+</sup> (Excellent)</b>	<b>9</b>
<b>Below 70% but not less than 60% (<math>\geq 60\%</math>, <math>&lt; 70\%</math>)</b>	<b>A (Very Good)</b>	<b>8</b>
<b>Below 60% but not less than 55% (<math>\geq 55\%</math>, <math>&lt; 60\%</math>)</b>	<b>B<sup>+</sup> (Good)</b>	<b>7</b>
<b>Below 55% but not less than 50% (<math>\geq 50\%</math>, <math>&lt; 55\%</math>)</b>	<b>B (above Average)</b>	<b>6</b>
<b>Below 50% but not less than 45% (<math>\geq 45\%</math>, <math>&lt; 50\%</math>)</b>	<b>C (Average)</b>	<b>5</b>
<b>Below 45% but not less than 40% (<math>\geq 40\%</math>, <math>&lt; 45\%</math>)</b>	<b>P (Pass)</b>	<b>4</b>
<b>Below 40% (<math>&lt; 40\%</math>)</b>	<b>F (FAIL)</b>	<b>0</b>

**10.3** A student obtaining F Grade in any Subject shall be considered 'failed' and will be required to reappear as 'Supplementary Candidate' in the End Semester Examination (SEE), as and when offered. In such cases, his Internal Marks (CIE Marks) in those Subject(s) will remain same as those he obtained earlier.

**10.4** A Letter Grade does not imply any specific % of Marks.

**10.5** In general, a student shall not be permitted to repeat any Subject/ Course (s) only for the sake of 'Grade Improvement' or 'SGPA/ CGPA Improvement'. However, he has to repeat all the Subjects/ Courses pertaining to that Semester, when he is detained (as listed in Items 8.10-8.11).

**10.6** A student earns Grade Point (GP) in each Subject/ Course, on the basis of the Letter Grade obtained by him in that Subject/ Course (excluding Mandatory non-credit Courses). Then the corresponding 'Credit Points' (CP) are computed by multiplying the Grade Point with Credits for that particular Subject/ Course.

**Credit Points (CP) = Grade Point (GP) x Credits ....  
For a Course**



**10.7** The Student passes the Subject/ Course only when he gets  $GP \geq 4$  (P Grade or above).

**10.8** The Semester Grade Point Average (SGPA) is calculated by dividing the Sum of Credit Points ( $\Sigma CP$ ) secured from ALL Subjects/ Courses registered in a Semester, by the Total Number of Credits registered during that Semester. SGPA is rounded off to TWO Decimal Places. SGPA is thus computed as

$$SGPA = \{ \sum_{i=1}^N C_i G_i \} / \{ \sum_{i=1}^N C_i \} \dots \text{For each Semester,}$$

where 'i' is the Subject indicator index (takes into account all Subjects in a Semester), 'N' is the no. of Subjects 'REGISTERED' for the Semester (as specifically required and listed under the Course Structure of the parent Department), is the no. of Credits allotted to the ith Subject, and represents the Grade Points (GP) corresponding to the Letter Grade awarded for that ith Subject.

**10.9** The Cumulative Grade Point Average (CGPA) is a measure of the overall cumulative performance of a student over all Semesters considered for registration. The CGPA is the ratio of the Total Credit Points secured by a student in ALL registered Courses in ALL Semesters, and the Total Number of Credits registered in ALL the Semesters. CGPA is rounded off to TWO Decimal Places. CGPA is thus computed from the I Year Second Semester onwards, at the end of each Semester, as per the formula

$$CGPA = \{ \sum_{j=1}^M C_j G_j \} / \{ \sum_{j=1}^M C_j \} \dots \text{for all } S \text{ Semesters registered (ie., upto and inclusive of } S \text{ Semesters, } S \geq 2),$$

where 'M' is the TOTAL no. of Subjects (as specifically required and listed under the Course Structure of the parent Department) the Student has 'REGISTERED' from the 1<sup>st</sup> Semester onwards upto and inclusive of the Semester S (obviously  $M > N$ ), 'j' is the Subject indicator index (takes into account all Subjects from 1 to S Semesters), is the no. of Credits allotted to the jth Subject, and represents the Grade Points (GP) corresponding to the Letter Grade awarded for that jth Subject. After registration and completion of I Year I Semester however, the SGPA

of that Semester itself may be taken as the CGPA, as there are no cumulative effects.

**10.10** For Merit Ranking or Comparison Purposes or any other listing, ONLY the 'ROUNDED OFF' values of the CGPAs will be used.

**10.11** For Calculations listed in Item 10.6 – 10.10, performance in failed Subjects/ Courses (securing F Grade) will also be taken into account, and the Credits of such Subjects/ Courses will also be included in the multiplications and summations. However, Mandatory Courses will not be taken into consideration.

#### **10.12 Passing Standards:**

**10.12.1** A student shall be declared successful or 'passed' in a Semester, only when he gets a  $SGPA \geq 5.00$  (at the end of that particular Semester); and a student shall be declared successful or 'passed' in the entire UGP, only when gets a  $CGPA \geq 5.00$ ; subject to the condition that he secures a  $GP \geq 4$  (P Grade or above) in every registered Subject/ Course in each Semester (during the entire UGP) for the Degree Award, as required.

**10.12.2** In spite of securing P Grade or above in some (or all) Subjects/ Courses in any Semester, if a Student receives a  $SGPA < 5.00$  and/ or  $CGPA < 5.00$  at the end of such a Semester, then he 'may be allowed' (on the 'specific recommendations' of the Head of the Department and subsequent approval from the Principal) -

(i) to go into the next subsequent Semester (subject to fulfilling all other attendance and academic requirements as listed under Items 7-8);

(ii) to 'improve his SGPA of such a Semester (and hence CGPA) to 5.00 or above', by reappearing for ONE or MORE (as per Student's choice) of the same Subject(s) / Course(s) in which he has secured P Grade(s) in that Semester, at the Supplementary Examinations to be held in the next subsequent Semester(s). In such cases, his Internal Marks (CIE Marks) in those Subject(s) will remain same as those he obtained earlier.

In these considerations, the newly secured Letter Grades will be recorded and taken into account for calculation of SGPA and CGPA, only if there is an improvement.

**10.12.3** A Student shall be declared successful or 'passed' in any Non-Credit Subject/ Course, if he secures a 'Satisfactory Participation Certificate' for that Mandatory Course.

**10.13** After the completion of each Semester, a Grade Card or Grade Sheet (or Transcript) shall be issued to all the Registered Students of that Semester, indicating the Letter Grades and Credits earned. It will show the details of the Courses Registered (Course Code, Title, No. of Credits, Grade Earned etc.), Credits earned, SGPA, and CGPA.

## **11.0 Declaration of Results**

**11.1** Computation of SGPA and CGPA are done using the procedure listed in 10.6 – 10.10.

**11.2** For Final % of Marks equivalent to the computed final CGPA, the following formula may be used ...

$$\% \text{ of Marks} = (\text{final CGPA} - 0.5) \times 10$$

## **12.0 Award of Degree**

**12.1** A Student who registers for all the specified Subjects/ Courses as listed in the Course Structure, satisfies all the Course Requirements, and passes all the examinations prescribed in the entire UG E&T Programme (UGP), and secures the required number of 192 Credits (with CGPA  $\geq 5.0$ ), within 8 Academic Years from the Date of Commencement of the First Academic Year, shall be declared to have 'QUALIFIED' for the Award of the B.Tech. Degree in the chosen Branch of Engineering as selected at the time of Admission.

**12.2** A Student who qualifies for the Award of the Degree as listed in Item 12.1, shall be placed in the following Classes ...

**12.3** Students with final CGPA (at the end of the UGP)  $\geq 8.00$ , and fulfilling the following conditions -

- (i) should have passed all the Subjects/Courses in 'FIRST APPEARANCE' within the first 4 Academic Years (or 8 Sequential Semesters) from the Date of Commencement of his First Academic Year,

- (ii) should have secured a CGPA  $\geq 8.00$ , at the end of each of the 8 Sequential Semesters, starting from the I Year I Semester onwards,

- (iii) should not have been detained or prevented from writing the End Semester Examinations in any Semester due to shortage of attendance or any other reason, shall be placed in 'FIRST CLASS with DISTINCTION'.

Students having final CGPA (at the end of UGP)  $\geq 8.00$ , but not fulfilling the above conditions shall be placed in 'FIRST CLASS'.

**12.4** Students with final CGPA (at the end of the UGP)  $\geq 6.50$  but  $< 8.00$ , shall be placed in 'FIRST CLASS'.

**12.5** Students with final CGPA (at the end of the UGP)  $\geq 5.50$  but  $< 6.50$ , shall be placed in 'SECOND CLASS'.

**12.6** All other Students who qualify for the Award of the Degree (as per Item 12.1), with final CGPA (at the end of the UGP)  $\geq 5.00$  but  $< 5.50$ , shall be placed in 'PASS CLASS'.

**12.7** A student with final CGPA (at the end of the UGP)  $< 5.00$  will not be eligible for the Award of the Degree.

**12.8** Students fulfilling the conditions listed under Item 12.3 alone will be eligible candidates for - 'University Rank' and 'Gold Medal' considerations.

## **13.0 Withholding of Results**

**13.1** If the student has not paid fees to University/ College at any stage, or has pending dues against his name due to any reason whatsoever, or if any case of indiscipline is pending against him, the result of the student may be withheld, and he will not be allowed to go into the next higher Semester. The Award or issue of the Degree may also be withheld in such cases.

**14.0 Transitory Regulations**

- 14.1** Student who has discontinued for any reason, or has been detained for want of attendance or lack of required credits as specified, or who has failed after having undergone the Degree Programme, may be considered eligible for readmission to the same Subjects/ Courses (or equivalent Subjects/ Courses, as the case may be), and same Professional Electives/ Open Electives (or from set/category of Electives or equivalents suggested, as the case may be) as and when they are offered (within the time-frame of 8 years from the Date of Commencement of his I Year I Semester).

**15.0 Student Transfers**

- 15.1** There shall be no Branch transfers after the completion of Admission Process.
- 15.2** There shall be no transfer among the Constituent Colleges and Units of Jawaharlal Nehru Technological University Hyderabad.

**16.0 Scope**

- i) Where the words “he”, “him”, “his”, occur in the write-up of regulations, they include “she”, “her”, “hers”.
- ii) Where the words “Subject” or “Subjects”, occur in these regulations, they also imply “Course” or “Courses”.
- iii) The Academic Regulations should be read as a whole, for the purpose of any interpretation.
- iv) In case of any doubt or ambiguity in the interpretation of the above rules, the decision of the Vice-Chancellor/ Principal is final.
- v) The College may change or amend the Academic Regulations, Course Structure or Syllabi at any time, and the changes or amendments made shall be applicable to all Students with effect from the dates notified by the College Authorities.

\* \* \* \* \*

**MALPRACTICES RULES**

	<b>Nature of Malpractices</b>	<b>Punishment</b>
	<b>If the candidate:</b>	
1 (a)	Possesses or keeps accessible in examination hall, any paper, note book, programmable calculators, Cell phones, pager, palm computers or any other form of material concerned with or related to the subject of the examination (theory or practical) in which he is appearing but has not made use of (material shall include any marks on the body of the candidate which can be used as an aid in the subject of the examination)	Expulsion from the examination hall and cancellation of the performance in that subject only.
1 (b)	Gives assistance or guidance or receives it from any other candidate orally or by any other body language methods or communicates through cell phones with any candidate or persons in or outside the exam hall in respect of any matter.	Expulsion from the examination hall and cancellation of the performance in that subject only of all the candidates involved. In case of an outsider, he will be handed over to the police and a case is registered against him.
2	Has copied in the examination hall from any paper, book, programmable calculators, palm computers or any other form of material relevant to the subject of the examination (theory or practical) in which the candidate is appearing.	Expulsion from the examination hall and cancellation of the performance in that subject and all other subjects the candidate has already appeared including practical examinations and project work and shall not be permitted to appear for the remaining examinations of the subjects of that Semester/year. The Hall Ticket of the candidate is to be cancelled.
3	Impersonates any other candidate in connection with the examination.	The candidate who has impersonated shall be expelled from examination hall. The candidate is also debarred and forfeits the seat. The performance of the original candidate who has been impersonated, shall be cancelled in all the subjects of the examination

		(including practicals and project work) already appeared and shall not be allowed to appear for examinations of the remaining subjects of that semester/year. The candidate is also debarred for two consecutive semesters from class work and all examinations. The continuation of the course by the candidate is subject to the academic regulations in connection with forfeiture of seat. If the imposter is an outsider, he will be handed over to the police and a case is registered against him.
4	Smuggles in the Answer book or additional sheet or takes out or arranges to send out the question paper during the examination or answer book or additional sheet, during or after the examination.	Expulsion from the examination hall and cancellation of performance in that subject and all the other subjects the candidate has already appeared including practical examinations and project work and shall not be permitted for the remaining examinations of the subjects of that semester/year. The candidate is also debarred for two consecutive semesters from class work and all examinations. The continuation of the course by the candidate is subject to the academic regulations in connection with forfeiture of seat.
5	Uses objectionable, abusive or offensive language in the answer paper or in letters to the examiners or writes to the examiner requesting him to award pass marks.	Cancellation of the performance in that subject.
6	Refuses to obey the orders of the Chief Superintendent / Assistant –Superintendent / any officer on duty or misbehaves or creates disturbance of any kind in and around the examination hall or organizes a walk out or instigates others to walk out, or threatens the officer-in charge or any person on duty in or outside the	In case of students of the college, they shall be expelled from examination halls and cancellation of their performance in that subject and all other subjects the candidate(s) has (have) already appeared and shall not be permitted to appear for the remaining examinations of the subjects of that semester/year. The candidates also are debarred and forfeit their seats. In case of outsiders, they will be handed over to

	examination hall of any injury to his person or to any of his relations whether by words, either spoken or written or by signs or by visible representation, assaults the officer-in-charge, or any person on duty in or outside the examination hall or any of his relations, or indulges in any other act of misconduct or mischief which result in damage to or destruction of property in the examination hall or any part of the College campus or engages in any other act which in the opinion of the officer on duty amounts to use of unfair means or misconduct or has the tendency to disrupt the orderly conduct of the examination.	the police and a police case is registered against them.
7	Leaves the exam hall taking away answer script or intentionally tears of the script or any part thereof inside or outside the examination hall.	Expulsion from the examination hall and cancellation of performance in that subject and all the other subjects the candidate has already appeared including practical examinations and project work and shall not be permitted for the remaining examinations of the subjects of that semester/year. The candidate is also debarred for two consecutive semesters from class work and all examinations. The continuation of the course by the candidate is subject to the academic regulations in connection with forfeiture of seat.
8	Possess any lethal weapon or firearm in the examination hall.	Expulsion from the examination hall and cancellation of the performance in that subject and all other subjects the candidate has already appeared including practical examinations and project work and shall not be permitted for the remaining examinations of the subjects of that semester/year. The candidate is also debarred and forfeits the seat.
9	If student of the college, who	Student of the colleges expulsion

	is not a candidate for the particular examination or any person not connected with the college indulges in any malpractice or improper conduct mentioned in clause 6 to 8.	from the examination hall and cancellation of the performance in that subject and all other subjects the candidate has already appeared including practical examinations and project work and shall not be permitted for the remaining examinations of the subjects of that semester/year. The candidate is also debarred and forfeits the seat. Person(s) who do not belong to the College will be handed over to police and, a 8police case will be registered against them.
10	Comes in a drunken condition to the examination hall.	Expulsion from the examination hall and cancellation of the performance in that subject and all other subjects the candidate has already appeared including practical examinations and project work and shall not be permitted for the remaining examinations of the subjects of that semester/year.
11	Copying detected on the basis of internal evidence, such as, during valuation or during special scrutiny.	Cancellation of the performance in that subject and all other subjects the candidate has appeared including practical examinations and project work of that semester / year examinations.
12	If any malpractice is detected which is not covered in the above clauses 1 to 11 shall be reported to the College / University for further action to award suitable punishment.	

**JNTUH COLLEGE OF ENGINEERING HYDERABAD  
(AUTONOMOUS)**

**CHEMICAL ENGINEERING**

**COURSE STRUCTURE**

(Applicable from the batch admitted during 2015-16 and onwards)

**I YEAR**

**I SEMESTER**

S.No.	Group	Subject	L	T	P	Credits
1	BS	Mathematics - I	4	1	0	4
2	BS	Engineering Physics	3	1	0	3
3	BS	Applied Chemistry	3	1	0	3
4	EAS	Computer Programming & Data Structures	4	1	0	4
5	EAS	Applied Mechanics	4	1	0	4
6	BS	Engineering Physics Lab	0	0	3	2
7	BS	Applied Chemistry Lab	0	0	3	2
8	EAS	Computer Programming & Data Structures Lab	0	0	3	2
		NSS/NCC/NSO				
		<b>Total Credits</b>				<b>24</b>

**I YEAR**

**II SEMESTER**

S.No.	Group	Subject	L	T	P	Credits
1	BS	Mathematics-II	3	1	0	3
2	EAS	Fundamentals of Electrical & Electronics Engineering	3	0	0	3
3	HSS	English	3	0	0	3
4	EAS	Engineering Graphics	3	0	3	4
5	EAS	Environmental Science	3	0	0	3
6	BS	Computational Mathematics	2	0	0	2
7	EAS	Engineering Workshop	0	0	3	2
8	HSS	English Language Communication Skills Lab	0	0	3	2
9	BS	Computational Mathematics Lab	0	0	3	2
		NSS/NCC/NSO				
		<b>Total Credits</b>				<b>24</b>

**JNTUH COLLEGE OF ENGINEERING HYDERABAD**

**I Year B.Tech. Chem. Engg. I-Sem**

**L T P C**  
**4 1 0 4**

**MATHEMATICS – I  
(Common to all Branches)**

**Pre Requisites: NIL**

**Objectives:**

- To train the students thoroughly in mathematical concepts of ordinary differential equations and their applications.
- To prepare students for lifelong learning and successful careers using mathematical Concepts of differential and integral calculus, ordinary differential equations and vector calculus.
- To develop the skill pertinent to the practice of the mathematical concepts including the students abilities to formulate and modeling the problems, to think creatively and to synthesize information.

**Outcomes:**

At the end of the course, the student will be able to:

- become familiar with the application of differential and integral calculus, ordinary differential equations and vector calculus to engineering problems.
- attain the abilities to use mathematical knowledge to analyze, formulate and solve problems in engineering applications.

**UNIT-I: Differential calculus (12 lectures)**

Rolle's Mean value Theorem – Lagrange's Mean Value Theorem – Cauchy's mean value Theorem – (all theorems without proof but with geometrical interpretations), verification of the Theorems and testing the applicability of these theorem to the given function.

Curve tracing – Equations given in Cartesian, polar and parametric forms.

Functions of several variables – Functional dependence- Jacobian- Maxima and Minima of functions of two variables with constraints and without constraints-Method of Lagrange multipliers.

**UNIT-II: Improper Integrals, Multiple Integration (12 lectures)**

Gamma and Beta Functions –Relation between them, their properties – evaluation of improper integrals using Gamma / Beta functions.

Multiple integrals – double and triple integrals – change of order of integration- change of variables (polar, cylindrical and spherical) .

Finding the area of a region using Double integration and volume of a region in space using triple integration.

### UNIT-III: Vector Calculus (12 lectures)

Vector Calculus: Scalar point function and vector point function, Gradient-Divergence- Curl and their related properties, - Laplacian operator, Line integral – Work done – Surface integrals –Volume integral. Green's Theorem, Stoke's theorem and Gauss's Divergence Theorems (Statement & their Verification). Solenoidal and irrotational vectors, Finding potential function.

### UNIT-IV: First Order Ordinary Differential Equations (10 lectures)

Linear and exact differential equations

Applications of first order differential equations – Newton's Law of cooling, Law of natural growth and decay, orthogonal trajectories and electrical circuits

### UNIT-V: Higher Order Ordinary Differential Equations (10 lectures)

Linear, homogeneous and non- homogeneous differential equations of second and higher order with constant coefficients. Non-homogeneous term of the type  $e^{ax}$ ,  $\sin ax$ ,  $\cos ax$ , and  $x^n$ ,  $e^{ax}V(x)$ ,  $x^n V(x)$ . Method of variation of parameters. Applications: Bending of beams, Electrical circuits and simple harmonic motion.

#### Text books:

- 1) HIGHER ENGINEERING MATHEMATICS BY B S GREWAL, KHANNA PUBLICATIONS.
- 2) ENGINEERING MATHEMATICS BY ERWIN KREYSZIG, WILEY PUBLICATIONS.
- 3) VECTOR ANALYSIS BY GHOSH & MAITY, NEW CENTRAL BOOK AGENCY.

#### References:

- 1) ENGINEERING MATHEMATICS BY SRIMANTAPAL & SUBODH C. BHUNIA, OXFORD UNIVERSITY PRESS.
- 2) ADVANCED ENGINEERING MATHEMATICS BY PETER V O'NEIL, CENGAGE LEARNING.

## JNTUH COLLEGE OF ENGINEERING HYDERABAD

I Year B.Tech. Chem. Engg. I-Sem

L	T	P	C
3	1	0	3

### ENGINEERING PHYSICS

**Prerequisites:** Nil

#### Course Objectives:

The course aims at making students to understand the basic concepts of Principles of Physics in a broader sense with a view to lay foundation for the various engineering courses. Today the need is to stress principles rather than specific procedures, to select areas of contemporary interest rather than of past interest, and to condition the student to the atmosphere of change he will encounter during his carrier. The main purpose of this course is to equip engineering undergraduates with an understanding of the scientific method, so that they may use the training beneficially in their higher pursuits. An earnest attempt is made in framing the course in this direction by blending careful presentation of fundamental concepts and methods of physics. The course begins with a detailed coverage of optics, which includes topics such as interference, diffraction, polarization of Light phenomenon. It then delves into discussion on Characterization of materials in terms of bonding, defects, Structures, X-ray diffraction, dielectric nature, Magnetic behavior, Superconducting nature, Nano size activity etc. The basic principles behind the Acoustics of good structures (Halls) are elucidated for easy understanding of complex concepts.

#### Outcomes:

The knowledge of Physics relevant to engineering is critical for converting ideas into technology. An understanding of Physics also helps engineers understand the working and limitations of existing devices and techniques, which eventually leads to new innovations and improvements. In the present course, the students can gain knowledge not only about one of the naturally main source of life on the earth planet i.e. sun rays and their phenomenon, but also of other artificial light source behavior and their phenomenon. Similarly, by studying other chapters, the student can gain the knowledge of the relation between the micro level physical properties of the materials and their macro level behavior thereby acquires the idea of making them use effectively in real time situation or in applications of interest.

**UNIT-I**

1. **Interference:** Superposition of Waves, Young's double slit experiment, Coherence, Interference in Thin films by Reflection, Newton's Rings.
2. **Diffraction:** Fresnel and Fraunhofer diffractions, Fraunhofer diffraction at a single slit and Double slit, Diffraction Grating, Resolving Power of a Grating.
3. **Polarization:** Introduction to polarization, Double Refraction, Nicol Prism, Quarter and Half wave plates

**UNIT-II**

4. **Bonding in Solids:** Ionic Bond, Covalent Bond, Metallic Bond, Hydrogen Bond, Vander- Waal's Bond and Calculation of Cohesive Energy for Ionic Solids.
5. **Crystallography and Crystal Structures:** Space Lattice, Unit Cell, Lattice parameters, Crystal Systems, Bravais Lattices, Miller Indices, Atomic Radius, Co-ordination Number and Packing Factor of SC, BCC, FCC, Diamond Structure(Cubic), Structures of NaCl, ZnS, CsCl, Crystal Planes and Directions, Inter Planar Spacing of Orthogonal Crystal Systems

**UNIT-III**

6. **Defects in Crystals:** Point Defects: Vacancies, Substitutional, Interstitial, Frenkel and Schottky Defects, Estimation of Frenkel and Schottky Defects, Edge and Screw dislocations (Qualitative Treatment), Burger's Vector.
7. **X-ray Diffraction:** Basic Principles of X-ray Diffraction, Bragg's Law, Laue Method, Powder Method, Applications of X-ray Diffraction.

**UNIT-IV**

8. **Dielectric Properties:** Basic definitions: Electric dipole, Dipole moment, Permittivity, Dielectric constant, Polarizability, Electric susceptibility, Displacement vector; Electronic Polarization, Ionic Polarization (Quantitative treatment) and Orientation Polarization (Qualitative treatment), Internal Fields in Solids, Clausius - Mossotti Equation, Piezo-electricity, Pyro- electricity and Ferro – electricity, Properties of ferro-electric materials.
9. **Magnetic Properties:** Basic definitions: Magnetic dipole, Magnetic moment, Magnetic Induction, Magnetic field intensity, Permeability, relative permeability, Intensity of Magnetization, susceptibility, Relation between  $\chi$  and  $\mu_r$ ; Origin of Magnetic Moment-Bohr Magneton, Classification of Dia, Para and Ferro Magnetic Materials on the basis of Magnetic Moment, Domain Theory of Ferro magnetism on the basis of Hysteresis Curve, Soft and Hard Magnetic Materials, Properties of Anti-Ferro and Ferri Magnetic Materials.

10. **Superconductivity:** Introduction of Superconductivity, Properties of Superconductors, Meissner Effect, BCS theory (Qualitative), Type-I and Type II Superconductors, Magnetic Levitation and Applications of Superconductors.

**UNIT-V**

11. **Acoustics of Buildings:** Basic Requirement of Acoustically Good Hall, Reverberation and Time of Reverberation, Sabine's Formula for Reverberation Time, Measurement of Absorption Coefficient of a material, Factors Effecting the Architectural Acoustics and their Remedies.
12. **Nanomaterials:** Origin of Nanotechnology, Nano Scale, Surface to Volume Ratio, Quantum Confinement, Bottom-up Fabrication: Sol-Gel, Precipitation, combustion Methods; Top-Down Fabrication: Chemical Vapor Deposition, Physical Vapor Deposition, Characterization Techniques (XRD, SEM & TEM) and Applications.

**Text books:**

1. Principles of Physics by Halliday, Resnick, Walker, Wiley India Pvt Ltd, 9<sup>th</sup> Edition.
2. Introduction to Solid State Physics by Charles Kittel, Wiley India Pvt Ltd, 7<sup>th</sup> Edition
3. Engineering Physics by R.K.GAUR & S.L.GUPTA, Dhanpat Rai Publications.
4. Solid State Physics by A J Dekker, MACMILLAN INDIA LTD.

**References:**

1. Modern Engineering Physics by Dr.K.Vijaya Kumar, Dr.S.Chandralingam, S.CHAND & COMPANY LTD
2. Applied Physics by P.K.Mittal, I K International Publishers
3. Applied Physics by P.K. Palanisamy :Scitech publishers
4. Introduction to Nanotechnology by Charles P.Poole, Jr.Frank J ownes, John Wiley & sons
5. Applied Physics for Engineers by P. Madusudana Rao, Academic Publishing Company
6. Engineering Physics by Sanjay D Jain, Girish G Sahasrbudha: University Press.



**JNTUH COLLEGE OF ENGINEERING HYDERABAD****I Year B.Tech. Chem. Engg. I-Sem**

L	T	P	C
3	1	0	3

**APPLIED CHEMISTRY****Prerequisites: Nil****Course objectives:**

The student is made to understand the basic concepts of chemistry. To explore the economically viable techniques developed for utilizing water resources and to provide the skills for analysis of water and materials. To inculcate the knowledge of corrosion to face thin burning issue of the Globe. The principle of the preparing and utilizing various polymers for varied applications.

**Outcomes:**

At the end of the course, the student will be able to:

- learn the concepts of electrochemistry, batteries. The principles and procedures for making varied polymers for different applications are well understood. The skills pertaining to water treatment and analysis will be inculcated.

**Unit-I: Water and its treatment**

Introduction – hardness of water – causes of hardness – types of hardness : temporary and permanent – expression and units of hardness – Estimation of hardness of water by complexometric method. Potable water and its specifications. Steps involved in treatment of potable water - Disinfection of potable water by chlorination and Ozonization. Boiler feed water and its treatment – Calgon conditioning – Phosphate conditioning - Colloidal conditioning – External treatment of water –ion-exchange processes. Desalination of water – Reverse osmosis. Numerical problems – Sewage water - COD, BOD definition and their significance. Treatment of sewage- Steps involved.

**Unit-II: Electrochemistry and corrosion**

**Electrochemistry** :Conductance and their interrelationship - Specific, equivalent and molar conductance . Ionic mobilities – Relationship between ionic conductance and ionic mobilities. Electro Chemical cells - electrode potential, standard electrode potential, types of electrodes – Standard hydrogen electrode, calomel and glass electrode. Nernst equation - electrochemical series and its applications.– Concept of concentration cell –Numerical problems.

**Corrosion**- Causes and effects of corrosion – theories of chemical and electrochemical corrosion - mechanism of electrochemical corrosion. Types of corrosion : Galvanic, water-line and pitting corrosion. Factors affecting rate of corrosion. Corrosion control methods – Cathodic protection - sacrificial anode and impressed current cathodic methods. Surface coatings – metallic coatings – methods of application of metallic coatings – Hot dipping , cementation, electroplating of copper, electro less plating of Nickel - Organic coatings: Paints – their constituents and functions.

**Unit-III: High Polymers:**

Definition – Classification of polymers with examples – Types of polymerisation – Chain growth (free radical addition mechanism), step growth polymerization, Plastics, fibres and elastomers - definition and characteristics. Plastics – thermoplastic and thermosetting plastics, compounding of plastics . Fibre reinforced plastics. Preparation, properties and applications of PVC, Teflon, Bakelite, Nylon 6:6 and terylene (Dacron); Rubber – Natural rubber , its processing and vulcanization. Elastomers: Preparation, properties and applications of Styrene butadiene, butyl and thiokol rubbers. Conducting polymers – Classification with examples; mechanism of conduction in trans-polyacetylene and applications of conducting polymers. Biodegradable polymers – concept and advantages - Polylactic acid and their applications.

**Unit-IV: Chemistry of Energy sources**

**Fuels** :Classification of fuels - characteristics of a good fuel . Solid fuels: Coal – Analysis of coal by proximate and ultimate methods. Liquid fuels- Petroleum and its refining. Characteristics and uses of petrol, diesel and kerosene. Synthetic petrol- Fischer-Tropsch's process. Cracking – thermal cracking and catalytic cracking. Fluid bed catalytic cracking, Knocking - octane and cetane numbers. Gaseous fuels – Composition, properties and uses of Natural gas, LPG and CNG .

**Combustion** – Definition, calorific value, HCV and LCV. Calculation of air quantity required for combustion of a fuel - Numerical problems.

**Alternate Energy sources** :Biodiesel - trans-esterification - advantages of biodiesel, fuel cells ( $H_2-O_2$  and Methanol  $-O_2$  fuel cell).

**Unit-V: Engineering Materials:**

**Cement** – Types of cements- Portland cement and its composition, Setting and hardening of cement. Special Cements – White cement, Water proof cement, high-alumina cement and high early-strength cement .

**Refractories** – Characteristics of a good refractory, classification with examples – refractoriness and refractoriness under load - causes for the failure of refractories.

**Abrasives:** Characteristics – Classification and applications of Diamond and Carborandum (SiC)

**Lubricants** – Characteristics of a good lubricant – mechanism of lubrication – thick film, thin film and extreme pressure lubrication. Classification of lubricants with examples. Properties– viscosity, cloud point, pour point, flash point, fire point and mechanical stability (determination and their significance).

#### Text Books:

1. Engineering Chemistry by P.C.Jain & M.Jain ; Dhanpat Rai Publishing Company (P) Ltd., New Delhi, (15<sup>th</sup> Edition , 2005).
2. Engineering Chemistry by B.Rama Devi & Ch.Venkata Ramana Reddy ; Cengage Learning , 2012.

#### Reference Books:

1. A Text Book of Engineering Chemistry by Shashi Chawla, Dhanpat Rai & Co., New Delhi.(3<sup>rd</sup> Edition, 2003).
2. Engineering Chemistry by Y. Bharathi Kumari and C. Jyotsna, VGS Booklinks, 2012.
3. Text book of Engineering Chemistry by C P Murthy, C V Agarwal and A. Naidu; B.S.Publications, 2006.
4. Engineering Chemistry by M. Thirumala Chary and E. Lakshminarayana, Sci tech. Publications Pvt. Ltd., Chennai 2012.
5. Engineering Chemistry by B.Sivasankar, Tata McGraw-Hill Publishing Company Ltd., New Delhi 2008.
6. A Text Book of Engineering Chemistry by S.S. Dara, S.Chand Publications, (10<sup>th</sup> Edition, 2007).

## JNTUH COLLEGE OF ENGINEERING HYDERABAD

I Year B.Tech. Chem. Engg. I-Sem

L	T	P	C
4	1	0	4

### COMPUTER PROGRAMMING & DATA STRUCTURES

#### Prerequisites:

There are no prerequisites for this course, except that anyone who wants to learn C should have analytical skills and logical reasoning.

#### Objectives:

1. This course starts from the basics of computers and program development.
2. It covers various concepts of C programming language
3. It introduces searching and sorting algorithms
4. It provides an understanding of data structures such as stacks and queues.

#### Outcomes:

At the end of the course, the student will be able to:

1. Develop C programs for computing and real life applications using basic elements like control statements, arrays, functions, pointers and strings; and data structures like stacks, queues and linked lists.
2. Implement searching and sorting algorithms

#### UNIT - I

**Introduction to Computers** – Computer Systems, Computing Environments, Computer Languages, Creating and running programs, Software Development Method, Algorithms, Pseudo code, flow charts, applying the software development method.

**Introduction to C Language** – Background, Simple C programs, Identifiers, Basic data types, Variables, Constants, Input / Output, Operators. Expressions, Precedence and Associativity, Expression Evaluation, Type conversions, Bit wise operators, Statements, Simple C Programming examples.

#### UNIT – II

**Statements** – if and switch statements, Repetition statements – while, for, do-while statements, Loop examples, other statements related to looping – break, continue, go to, Simple C Programming examples.

**Designing Structured Programs-** Functions, basics, user defined functions, inter function communication, Scope, Storage classes-auto,

register, static, extern, scope rules, type qualifiers, recursion- recursive functions, Preprocessor commands, example C programs

### UNIT – III

**Arrays and Strings** – Concepts, using arrays in C, inter function communication, array applications, two – dimensional arrays, multidimensional arrays, C program examples. Concepts, C Strings, String Input / Output functions, arrays of strings, string manipulation functions, string / data conversion, C program examples.

**Pointers** – Introduction (Basic Concepts), Pointers for inter function communication, pointers to pointers, compatibility, memory allocation functions, array of pointers, programming applications, pointers to void, pointers to functions, command –line arguments.

### UNIT - IV

**Derived types** – Structures – Declaration, definition and initialization of structures, accessing structures, nested structures, arrays of structures, structures and functions, pointers to structures, self referential structures, unions, typedef, bit fields, enumerated types, C programming examples.

**Input and Output** – Concept of a file, streams, standard input / output functions, formatted input / output functions, text files and binary files, file input / output operations, file status functions (error handling), C program examples.

### UNIT – V

**Sorting and Searching** selection sort, bubble sort, insertion sort, linear and binary search methods.

**Data Structures** – Introduction to Data Structures, abstract data types, Linear list – singly linked list implementation, insertion, deletion and searching operations on linear list, Stacks-Operations, array and linked representations of stacks, stack applications, Queues-operations, array and linked representations.

### TEXT BOOKS:

1. C Programming & Data Structures, B.A.Forouzan and R.F. Gilberg, Third Edition, Cengage Learning.
2. Problem Solving and Program Design in C, J.R. Hanly and E.B. Koffman, Fifth Edition, Pearson Education.
3. The C Programming Language, B.W. Kernighan and Dennis M.Ritchie, PHI/Pearson Education

### REFERENCES:

1. C & Data structures – P. Padmanabham, Third Edition, B.S. Publications.
2. C Programming with problem solving, J.A. Jones & K. Harrow, dreamtech Press
3. Programming in C – Stephen G. Kochan, III Edition, Pearson Educaion.
4. C for Engineers and Scientists, H.Cheng, Mc.Graw-Hill International Edition
5. Data Structures using C – A.M.Tanenbaum, Y.Langsam, and M.J. Augenstein, Pearson Education / PHI
6. C Programming & Data Structures,E.Balagurusamy,TMH.
7. C Programming & Data Structures, P. Dey, M Ghosh R Thereja, Oxford University Press
8. C & Data structures – E V Prasad and N B Venkateswarlu, S.Chand & Co.

**JNTUH COLLEGE OF ENGINEERING HYDERABAD****I Year B.Tech. Chem. Engg. I-Sem**

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4	1	0	4

**APPLIED MECHANICS****Prerequisites: Nil****Objectives:**

- The study of Engineering Mechanics is very vital to the curriculum of Engineering Studies
- The objective of this course is to build up and enhance the knowledge of mechanics studied in the physics in the application orientation for engineering problems. This course enables the students to take up the further courses in Engineering orientation for respective branches

**Outcomes:**

In applied mechanic Program, the student will be given maximum flexibility in pursuing an area of special interest like

- Dynamics and Vibrations -its interfaces to materials and structures for understanding on applications. Towards characteristics of structures and study on material and structure
- Applied Mathematics and Numerical Methods
- Experimental Mechanics and Materials
- Strength of materials & structural analysis

After completing the course student will have the basic knowledge on material and structures at analysis level. This will also integrate the science, engineering and mathematical concept for student understanding.

**UNIT – I**

**INTRODUCTION OF ENGINEERING. MECHANICS** – Basic concepts System of Forces- Coplanar Forces – Components in Space – Resultant-Moment of Forces and its Application – Couples and Resultant of Force System - Equilibrium of System of Forces- Free body diagrams-Direction of Force Equations of Equilibrium of Coplanar Systems and Spatial Systems – Vector cross product- Support reactions different beams for different types of loading – concentrated, uniformly distributed and uniformly varying loading .

**UNIT – II**

**FRICTION:** Types of friction – Limiting friction – Laws of Friction – static and Dynamic Frictions – Angle of Friction –Cone of limiting friction–

Friction of wedge, block and Ladder – Screw jack – Differential screw jack - Motion of Bodies.

**UNIT – III**

**CENTROID AND CENTER OF GRAVITY:** Centroids – Theorem of Pappus- Centroids of Composite figures – Centre of Gravity of Bodies - Area moment of Inertia: – polar Moment of Inertia – Transfer – Theorems - Moments of Inertia of Composite Figures.

**MOMENT OF INERTIA:** Moment of Inertia of Areas and Masses - Transfer Formula for Moments of Inertia - Moment of inertia of composite areas and masses.

**UNIT – IV**

**KINEMATICS:** Introduction – Rectilinear motion – Motion with uniform and variable acceleration – Curvilinear motion – Components of motion – Circular motion – Projectiles- Instantaneous centre.

**KINETICS:** Kinetics of a particle – D'Alembert's principle – Motion in a curved path – work, energy and power. Principle of conservation of energy – Kinetics of a rigid body in translation, rotation – work done – Principle of work-energy – Impulse-momentum.

**UNIT – V**

**MECHANICAL VIBRATIONS:** Definitions, Concepts – Simple Harmonic motion – free vibrations – Simple, compound and Torsional pendulums- Numerical problems.

**VIRTUAL WORK:** Theory of virtual work – Applications.

**TEXT BOOKS:**

1. Engineering Mechanics by shames & Rao - Pearson Education.
2. Engineering Mechanics by M.V. Seshagiri rao and Durgaih; University Press.
3. Engineering Mechanics – B. Bhattacharya - Oxford University Publications.

**REFERENCES:**

1. Engineering Mechanics (Statics and Dynamics) by Hibbler; Pearson Education.
2. Engineering Mechanics by Fedrinand L. Singer – Harper Collings Publishers.
3. Engineering Mechanics by A. K. Tayal, Umesh Publication.
4. Engineering Mechanics – G. S. Sawhney, Printice Hall of India.
5. A text book of engineering mechanics by R. K. Bansal; Laxmi publications.
6. Engineering Mechanics by R. S. Khurmi ; S. Chand & Co.

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**ENGINEERING PHYSICS LAB****LIST OF EXPERIMENTS**

1. Dispersive power of the material of a prism –Spectrometer
2. Determination of wavelengths of a source-Diffraction Grating.
3. Newton's Rings-Radius of curvature of Plano convex lens.
4. Time constant of an R-C Circuit.
5. Magnetic field along the axis of current carrying coil-Stewart and Gee's method.
6. Bending Losses of Fibers & Evaluation of numerical aperture of given fiber.
7. Energy gap of material of PN- junction.
8. Torsional pendulum.
9. Determination of frequency of A.C Mains-Sonometer.
10. Diffraction grating using single slit- Laser source

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**APPLIED CHEMISTRY LAB****LIST OF EXPERIMENTS**

- 1) Estimation of ferrous iron by Permanganometry.
- 2) Estimation of ferric iron by Dichrometry,
- 3) Estimation of copper by Iodometry
- 4) Estimation of  $\text{Fe}^{2+}$  &  $\text{Fe}^{3+}$  by dichrometry.
- 5) Estimation of hardness of water by Complexometry using EDTA,
- 6) Estimation of copper by Complexometry using EDTA.
- 7) Estimation of alkalinity of water.
- 8) Estimation of Permanent hardness by EDTA
- 9) Preparation of Thikol rubber
- 10) Estimation of iron in cement by Colorimetry;  $\text{KMnO}_4$
- 11) Estimation of Mn in  $\text{KMnO}_4$  by Colorimetry;
- 12) Estimation of HCl in a given solution by pH - metry.
- 13) Estimation of HCl in acid mixture by conductometry.
- 14) Estimation of  $\text{Fe}^{2+}$  by Potentiometry.

**Recommended Books:**

1. Inorganic Quantitative Analysis by A.I.Vogel, ELBS Publications (2007).
2. Laboratory Manual of Engineering Chemistry by Y.Bharathi Kumari & Jyotsna C, VGS Booklinks, Vijayawada, 2009.
3. College Practical Chemistry by V.K.Ahulwalia., Narosa Publications Ltd., New Delhi (2007)

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**COMPUTER PROGRAMMING & DATA STRUCTURES LAB****Week 1:**

1. Write a C program to find the sum of individual digits of a positive integer.
2. Fibonacci sequence is defined as follows: the first and second terms in the sequence are 0 and 1.  
Subsequent terms are found by adding the preceding two terms in the sequence. Write a C program to generate the first n terms of the sequence.
3. Write a C program to generate all the prime numbers between 1 and n, where n is a value supplied by the user.
4. Write a C program to find the roots of a quadratic equation.

**Week 2:**

5. Write a C program to find the factorial of a given integer.
6. Write a C program to find the GCD (greatest common divisor) of two given integers.
7. Write a C program to solve Towers of Hanoi problem.
8. Write a C program, which takes two integer operands and one operator from the user, performs the operation and then prints the result. (Consider the operators +, -, \*, /, % and use Switch Statement)

**Week 3:**

9. Write a C program to find both the largest and smallest number in a list of integers.
10. Write a C program that uses functions to perform the following:
  - i) Addition of Two Matrices
  - ii) Multiplication of Two Matrices

**Week 4:**

11. Write a C program that uses functions to perform the following operations:
  - i) To insert a sub-string in to a given main string from a given position.
  - ii) To delete n Characters from a given position in a given string.
12. Write a C program to determine if the given string is a palindrome or not

13. Write a C program that displays the position or index in the string S where the string T begins, or – 1 if S doesn't contain T.
14. Write a C program to count the lines, words and characters in a given text.

**Week 5:**

15. Write a C program to generate Pascal's triangle.
16. Write a C program to construct a pyramid of numbers.
17. Write a C program to read in two numbers, x and n, and then compute the sum of this geometric progression:  

$$1+x+x^2+x^3+\dots+x^n$$
 For example: if n is 3 and x is 5, then the program computes 1+5+25+125.  
 Print x, n, the sum  
 Perform error checking. For example, the formula does not make sense for negative exponents – if n is less than 0. Have your program print an error message if n<0, then go back and read in the next pair of numbers without computing the sum. Are any values of x also illegal? If so, test for them too.

**Week 6:**

18. 2's complement of a number is obtained by scanning it from right to left and complementing all the bits after the first appearance of a 1. Thus 2's complement of 11100 is 00100. Write a C program to find the 2's complement of a binary number.
19. Write a C program to convert a Roman numeral to its decimal equivalent.

**Week 7:**

20. Write a C program that uses functions to perform the following operations:
  - i) Reading a complex number
  - ii) Writing a complex number
  - iii) Addition of two complex numbers
  - iv) Multiplication of two complex numbers
 (Note: represent complex number using a structure.)

**Week 8:**

21. i) Write a C program which copies one file to another.  
 ii) Write a C program to reverse the first n characters in a file.  
 (Note: The file name and n are specified on the command line.)
22. i) Write a C program to display the contents of a file.  
 ii) Write a C program to merge two files into a third file (i.e., the contents of the first file followed by those of the second are put in the third file)

**JNTUH COLLEGE OF ENGINEERING HYDERABAD****Week 9:**

23. Write a C program that uses functions to perform the following operations on singly linked list.:

- i) Creation                      ii) Insertion                      iii) Deletion  
iv) Traversal

**Week 10:**

24. Write C programs that implement stack (its operations) using

- i) Arrays                      ii) Pointers

25. Write C programs that implement Queue (its operations) using

- i) Arrays                      ii) Pointers

**Week 11:**

26. Write a C program that implements the following sorting methods to sort a given list of integers in ascending order

- i) Bubble sort                      ii) Selection sort

**Week 12:**

27. Write C programs that use both recursive and non recursive functions to perform the following searching operations for a Key value in a given list of integers:

- i) Linear search                      ii) Binary search

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**MATHEMATICS – II**  
**(Common to all Branches)**

**Pre Requisites:** NIL

**Objectives:**

- Our emphasis will be more on conceptual understanding and application of Fourier series, Fourier, Z and Laplace transforms and solution of partial differential equations.

**Outcomes:**

At the end of the course, the student will be able to:

- gains the knowledge to tackle the engineering problems using the concepts of Fourier series, various transforms and partial differential equations.

**UNIT–I: Linear ODE with variable coefficients and series solutions**  
**(8 lectures)**

Equations reducible to constant coefficients-Cauchy's and Legendre's differential equations. Motivation for series solutions, Ordinary point and Regular singular point of a differential equation, Transformation of non-zero singular point to zero singular point. Series solutions to differential equations around zero, Frobenius Method about zero.

**Unit-II: Special Functions** **(8 lectures)**

Bessel's Differential equation, Bessel functions properties: – Recurrence relations, Orthogonality, Generating function, Trigonometric expansions involving Bessel functions.

**UNIT–III: Laplace Transform** **(8 lectures)**

Definition of Integral transform. Domain of the function and Kernel for the Laplace transforms, Laplace transform of standard functions, first shifting Theorem, Laplace transform of functions when they are multiplied or divided by "t". Laplace transforms of derivatives and integrals of functions. – Unit step function – second shifting theorem – Dirac's delta function, Periodic function – Inverse Laplace transform by Partial fractions( Heaviside method) Inverse Laplace transforms of functions when they are multiplied or divided by "s", Inverse Laplace Transforms of

derivatives and integrals of functions, Convolution theorem-solving differential equations by Laplace transforms

#### **UNIT – IV: Fourier series and Fourier Transforms (8 lectures)**

Definition of periodic function. Fourier expansion of periodic functions in a given interval of length,  $2\pi$ , Determination of Fourier coefficients – Fourier series of even and odd functions – Fourier series in an arbitrary interval – even and odd periodic continuation – Half-range Fourier sine and cosine expansions.

Fourier integral theorem – Fourier sine and cosine integrals. Fourier transforms – Fourier sine and cosine transforms – properties – inverse transforms – Finite Fourier transforms.

#### **UNIT-V: Partial Differential Equations (10 lectures)**

Introduction and Formation of partial differential equation by elimination of arbitrary constants and arbitrary functions, solutions of first order linear (Lagrange) equation and non-linear equations (Charpit's method).

Method of separation of variables for second order equations. Applications of Partial differential equations- one dimensional wave equation., Heat equation.

#### **Text books:**

- 1) HIGHER ENGINEERING MATHEMATICS BY B S GREWAL, KHANNA PUBLICATIONS.
- 2) ENGINEERING MATHEMATICS BY ERWIN KREYSZIG, WILEY PUBLICATIONS

#### **References:**

- 1) ENGINEERING MATHEMATICS BY SRIMANTAPAL & SUBODH C. BHUNIA, OXFORD UNIVERSITY PRESS.
- 2) ADVANCED ENGINEERING MATHEMATICS BY PETER V O'NEIL, CENGAGE LEARNING

## **JNTUH COLLEGE OF ENGINEERING HYDERABAD**

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### **FUNDAMENTALS OF ELECTRICAL AND ELECTRONICS ENGINEERING**

**Prerequisites: NIL**

#### **Objectives:**

- To introduce the concept of electrical circuits and its components.
- To introduce the characteristics of various electronic devices.
- To impart the knowledge of various configurations, characteristics and applications of electrical & electronic components.

#### **Outcomes:**

At the end of the course, the student will be able to:

- To analyze and solve electrical circuits using network laws and theorems.
- To design & analyse various circuits using electronic components viz. diodes, transistors & other special purpose devices.

#### **UNIT- I ELECTRICAL and SINGLE PHASE AC CIRCUITS**

**Electrical Circuits** - R-L-C Parameters, Voltage and Current Independent and Dependent Sources, Source Transformation – V–I relationship for Passive elements, Kirchoff's Laws, Network reduction techniques – series, parallel, series parallel, star-to-delta, delta-to-star transformation,

**Single Phase AC Circuits** - R.M.S. and Average values, Form Factor, steady state analysis of series, Parallel and Series parallel Combinations of R, L and C with Sinusoidal excitation, concept of reactance, Impedance, Susceptance and Admittance – phase and phase difference, Concept of Power Factor, j-notation.

#### **UNIT- II RESONANCE and NETWORK THEOREMS**

**Resonance** – Series resonance and Parallel resonance circuits, concept of bandwidth and Q factor, Locus Diagrams for RL, RC and RLC Combinations for Various Parameters.

**Network Theorems** - Thevenin's, Norton's, Maximum Power Transfer, Superposition, Reciprocity theorems for DC and AC excitations.



**UNIT- III P-N JUNCTION DIODE & DIODE CIRCUITS**

**P-N Junction Diode** - Diode equation, Energy Band diagram, Volt-Ampere characteristic, Temperature dependence, Static and dynamic resistances, Equivalent circuit, Load line analysis, Diffusion and Transition Capacitances.

**Rectifiers and Filters** - The P-N junction as a rectifier - A Half Wave Rectifier, Ripple Factor, Full Wave Rectifier, Bridge Rectifier, Harmonic components in Rectifier Circuits, Filters – Inductor Filters, Capacitor Filters, L- section Filters,  $\pi$ - section Filters.

**UNIT- IV BIPOLAR JUNCTION TRANSISTOR**

**Bipolar Junction Transistor (BJT)** - Construction, Principle of Operation, Symbol, Amplifying Action, Common Emitter, Common Base and Common Collector configurations.

**Transistor Configurations** - BJT modeling, Hybrid model, Determination of h-parameters from transistor characteristics, Analysis of CE, CB and CC configurations using h-parameters, Comparison of CE, CB and CC configurations.

**UNIT- V JUNCTION FIELD EFFECT TRANSISTOR & SPECIAL PURPOSE DEVICES:**

**Junction Field Effect Transistor** - Construction, Principle of Operation, Symbol, Pinch-Off Voltage, Volt-Ampere Characteristic, Comparison of BJT and JFET.

**Special Purpose Devices** - Breakdown Mechanisms in Semi Conductor Diodes, Zener diode characteristics, Use of Zener diode as simple regulator, Principle of operation and Characteristics of Tunnel Diode (With the help of Energy band diagram) and Varactor Diode, Principle of Operation of SCR.

**TEXT BOOKS:**

1. Electronic Devices and Circuits – R.L. Boylestad and Louis Nashelsky, PEI/PHI, 9<sup>th</sup> Ed, 2006.
2. Millman's Electronic Devices and Circuits – J.Millman and C.C.Halkias, Satyabratajit, TMH, 2/e, 1998.
3. Engineering circuit analysis- by William Hayt and Jack E. Kemmerly, Mc Graw Hill Company, 6<sup>th</sup> edition.

**REFERENCES:**

1. Introduction to Electronic Devices and Circuits-Rober T. Paynter, Pearson Education.

2. Electronic Devices and Circuits - K. Lal Kishore, B.S. Publications, 2<sup>nd</sup> Edition, 2005.
3. Electronic Devices and Circuits – Anil K. Maini, Varsha Agarwal – Wiley India Pvt. Ltd. 1/e 2009.
4. Linear circuit analysis (time domain phasor and Laplace transform approaches)- 2<sup>nd</sup> edition by Raymond A. DeCarlo and Pen-Min-Lin, Oxford University Press-2004.
5. Network Theory by N.C.Jagan & C.Lakshminarayana, B.S. Publications.
6. Network Theory by Sudhakar, Shyam Mohan Palli, TMH.

## JNTUH COLLEGE OF ENGINEERING HYDERABAD

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## ENGLISH

## 1. INTRODUCTION:

In view of the growing importance of English as a tool for global communication and the consequent emphasis on training students to acquire communicative competence, the syllabus has been designed to develop linguistic and communicative competencies of Engineering students. The prescribed books and the exercises are meant to serve broadly as students' handbooks.

In the English classes, the focus should be on the skills of reading, writing, listening and speaking and for this the teachers should use the text prescribed for detailed study. For example, the students should be encouraged to read the texts/selected paragraphs silently. The teachers can ask comprehension questions to stimulate discussion and based on the discussions students can be made to write short paragraphs/essays etc.

The text for non-detailed study is for extensive reading/reading for pleasure. Hence, it is suggested that they read the topics selected for discussion on their own in the class. The time should be utilized for working out the exercises given after each section, as also for supplementing the exercises with authentic materials of a similar kind for example, from newspaper articles, advertisements, promotional material, etc. *However, the stress in this syllabus is on skill development, fostering ideas and practice of language skills.*

## 2. OBJECTIVES:

- To improve the language proficiency of the students in English with emphasis on LSRW skills.
- To equip the students to study academic subjects more effectively using the theoretical and practical components of the English syllabus.
- To develop the study skills and communication skills in formal and informal situations.

## LEARNING OUTCOMES:

- Use of English Language - written and spoken.
- Enrichment of comprehension and fluency
- Gaining confidence in using language in verbal situations.

## SYLLABUS:

## Listening Skills:

## Objectives

- To enable students develop their listening skills so that they may appreciate the role in the LSRW skills approach to language and improve their pronunciation
- To equip students with necessary training in listening, so that they can comprehend the speech of people of different backgrounds and regions

*Students should be given practice in listening to the sounds of the language, to be able to recognize them, to distinguish between them, to mark stress and recognize and use the right intonation in sentences.*

- Listening for general content
- Listening to fill up information
- Intensive listening
- Listening for specific information

## Speaking Skills:

## Objectives

- To make students aware of the role of speaking in English and its contribution to their success.
- To enable students express themselves fluently and appropriately in social and professional contexts.
  - Oral practice
  - Describing objects/situations/people
  - Role play – Individual/Group activities (Using exercises from the five units of the prescribed text: **Skills Annexe-Functional English for Success**)
  - Just A Minute (JAM) Sessions.

## Reading Skills:

### Objectives

1. To develop an awareness in the students about the significance of silent reading and comprehension.
2. To develop the ability of students to guess the meanings of words from context and grasp the overall message of the text, draw inferences, etc.
  - Skimming the text
  - Understanding the gist of an argument
  - Identifying the topic sentence
  - Scanning
  - Inferring lexical and contextual meaning
  - Understanding discourse features
  - Recognizing coherence/sequencing of sentences

**NOTE:** *The students will be trained in reading skills using the prescribed text for detailed study.*

*They will be examined in reading and answering questions using 'unseen' passages which may be taken from authentic texts, such as magazines/newspaper articles.*

## Writing Skills:

### Objectives

1. To develop an awareness in the students about writing as an exact and formal skill
2. To equip them with the components of different forms of writing, beginning with the lower order ones.
  - Writing sentences
  - Use of appropriate vocabulary
  - Paragraph writing
  - Coherence and cohesiveness
  - Narration / description
  - Note Making
  - Formal and informal letter writing
  - Describing graphs using expressions of comparison

## TEXTBOOKS PRESCRIBED:

In order to improve the proficiency of the student in the acquisition of the four skills mentioned above, the following texts and course content, divided into **Five Units**, are prescribed:

**For Detailed study: First Textbook: “Skills Annexe -Functional English for Success”,** Published by Orient Black Swan, Hyderabad

### For Non-detailed study

**Second Textbook “Epitome of Wisdom”,** Published by Maruthi Publications, Hyderabad.

- The course content and study material is divided into **Five Units**.

## Unit –I

1. Chapter entitled ‘**Wit and Humour**’ from ‘**Skills Annexe**’ - **Functional English for Success**, Published by Orient Black Swan, Hyderabad
2. Chapter entitled ‘**Mokshagundam Visvesvaraya**’ from “**Epitome of Wisdom**”, Published by Maruthi Publications, Hyderabad.

L - Listening for Sounds, Stress and Intonation

S - Greeting and Taking Leave, Introducing Oneself and Others (Formal and Informal Situations)

R - Reading for Subject/ Theme- **The Palm Islands** from Epitome of Wisdom is for **Reading Comprehension**

W - Writing Paragraphs

G - Types of Nouns and Pronouns

V - Homonyms, Homophones & Homographs

## Unit –II

1. Chapter entitled “**Cyber Age**” from “**Skills Annexe -Functional English for Success**” Published by Orient Black Swan, Hyderabad.
2. **Report Writing (First & Second Textbooks)**

L - Listening for themes and facts

S - Apologizing, interrupting, requesting and making polite conversation

R- Reading for theme and gist- The 1 Thing Every Business Executive Must Understand about Social Media by Dave Kerpen from Skills Annexe is for **Reading Comprehension**

W - Describing people, places, objects, events

G - Verb forms

V - Noun, Verb, Adjective and Adverb

**Unit –III**

- Chapter entitled '**Risk Management**' from "**Skills Annexe - Functional English for Success**" Published by Orient Black Swan, Hyderabad
- Chapter entitled '**Leela's Friend**' by R.K. Narayan from "**Epitome of Wisdom**", Published by Maruthi Publications, Hyderabad
  - L - Listening for main points and sub-points for note taking
  - S - Giving instructions and directions; Speaking of hypothetical situations
  - R - Reading for details- *Sivakasi: Who to Blame for the Frequent Fire Accidents in India's Largest Fireworks Industry Hub?* by Amrutha Gayathri from Skills Annexe & *Forensic Science* from Epitome of Wisdom are for **Reading Comprehension**
  - W - Note-making, Information transfer, Punctuation
  - G - Present tense
  - V - Synonyms and Antonyms

**Unit –IV**

- Letter Writing – Writing formal letters, letter of application along with curriculum vitae (First & Second Textbooks)**
- Chapter entitled '**The Last Leaf**' from "**Epitome of Wisdom**", Published by Maruthi Publications, Hyderabad
  - L - Listening for specific details and information
  - S - Narrating, expressing opinions and telephone interactions
  - R - Reading for specific details and information- *What I Cherish Most* by V. S. Srinivasa Sastri from Skills Annexe & *Choose How to Start Your Day* from Epitome of Wisdom are for **Reading Comprehension**
  - W - Writing e-mails
  - G - Past and Future tenses
  - V - Vocabulary - Idioms and Phrasal verbs

**Unit –V**

- Chapter entitled '**Sports and Health**' from "**Skills Annexe - Functional English for Success**" Published by Orient Black Swan, Hyderabad
- Chapter entitled '**The Convocation Speech**' by N.R. Narayanmurthy' from "**Epitome of Wisdom**", Published by Maruthi Publications, Hyderabad
  - L - Critical Listening and Listening for speaker's tone/ attitude
  - S - Group discussion and Making presentations
  - R - Critical reading, reading for reference - *Benefits of Physical Activity* from Skills Annexe & *What is meant by*

*Entrepreneurship?* from Epitome of Wisdom are for **Reading Comprehension**

- W - Project proposals; Project Reports and Research Papers
- G - Adjectives, Prepositions and Concord
- V - Collocations and Technical vocabulary, Using words appropriately

✚ **Exercises from the texts not prescribed shall be used for classroom tasks.**

**REFERENCES:**

- Effective Technical Communication, M Ashraf Rizvi, Tata Mc Graw – Hill.
- Murphy's English Grammar with CD, Murphy, Cambridge University Press.
- Contemporary English Grammar Structures and Composition by David Green, MacMillan Publishers, New Delhi. 2010.
- Technical Communication, Meenakshi Raman, Oxford University Press
- Practical English Usage, Michael Swan, Oxford University Press
- Innovate with English: A Course in English for Engineering Students, edited by T Samson, Foundation Books.
- English Grammar Practice, Raj N Bakshi, Orient Longman.
- Technical Communication by Daniel Riordan. 2011. Cengage Publications. New Delhi.
- Handbook of English Grammar & Usage, Mark Lester and Larry Beason, Tata Mc Graw –Hill.
- Spoken English, R.K. Bansal & JB Harrison, Orient Longman.
- Grammar Games, Renuvolcuri Mario, Cambridge University Press.
- Everyday Dialogues in English, Robert J. Dixon, Prentice Hall India Pvt Ltd.,
- ABC of Common Errors Nigel D Turton, Mac Millan Publishers.
- Basic Vocabulary Edgar Thorpe & Showick Thorpe, Pearson Education
- An Interactive Grammar of Modern English, Shivendra K. Verma and Hemlatha Nagarajan, Frank Bros & CO

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**JNTUH COLLEGE OF ENGINEERING HYDERABAD****I Year B.Tech. Chem. Engg. II-Sem**

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**ENGINEERING GRAPHICS****Pre-requisites: Nil****Course objectives:**

- To provide basic concepts in engineering drawing.
- To impart knowledge about standard principles of orthographic projection of objects.
- To draw sectional views and pictorial views of solids.

**Outcomes:**

At the end of the course, the student will be able to:

- Preparing working drawings to communicate the ideas and information.
- Read, understand and interpret engineering drawings.

**UNIT – I****INTRODUCTION TO ENGINEERING DRAWING :**

Principles of Engineering Graphics and their Significance, Conic Sections including the Rectangular Hyperbola – General method only. Cycloid, Epicycloid and Hypocycloid, Involute. Scales – Plain, Diagonal and Vernier Scales.

**UNIT- II****ORTHOGRAPHIC PROJECTIONS:**

Principles of Orthographic Projections – Conventions – Projections of Points and Lines  
Projections of Plane regular geometric figures.—Auxiliary Planes.

**UNIT – III**

Projections of Regular Solids – Auxiliary Views.

**UNIT – IV**

Sections or Sectional views of Right Regular Solids – Prism, Cylinder, Pyramid, Cone – Auxiliary views – Sections of Sphere.  
Development of Surfaces of Right Regular Solids – Prism, Cylinder, Pyramid and Cone

**UNIT – V****ISOMETRIC PROJECTIONS :**

Principles of Isometric Projection – Isometric Scale – Isometric Views – Conventions – Isometric Views of Lines, Plane Figures, Simple and Compound Solids – Isometric Projection of objects having non-isometric lines. Isometric Projection of Spherical Parts.

Conversion of Isometric Views to Orthographic Views and Vice-versa – Conventions

Auto CAD: Basic principles only

**TEXT BOOKS:**

1. Engineering Drawing N.D. Bhatt / Charotar
2. Engineering Drawing and Graphics Rane and Shah/ Pearson Edu.

**REFERENCE BOOKS:**

1. A Text Book of Engineering Drawing / Dhawan R K / S. Chand
2. Engineering Graphics With Auto CAD / James D Bethune / Pearson Edu.
3. Engineering Graphics / K R Mohan / Dhanpat Rai.
4. Text book on Engineering Drawing / KL Narayana/ P Kannaih/ Scitech

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**ENVIRONMENTAL SCIENCE****Prerequisites : NIL****Objectives:**

- Creating the awareness about environmental problems among students.
- Imparting basic knowledge about the environment and its allied problems.
- Developing an attitude of concern for the environment.
- Motivating students to participate in environment protection and environment improvement.

**Outcomes:**

At the end of the course, it is expected that students will be able to:

- Identify and analyze environmental problems as well as the risks associated with these problems
- Understand what it is to be a steward in the environment
- Studying how to live their lives in a more sustainable manner

**UNIT- I****MULTIDISCIPLINARY NATURE OF ENVIRONMENTAL STUDIES:**

Definition, Scope and Importance – Need for Public Awareness.

**NATURAL RESOURCES** : Renewable and non-renewable resources – Natural resources and associated problems – Forest resources – Use and over – exploitation, deforestation, case studies – Timber extraction – Mining, dams and other effects on forest and tribal people – Water resources – Use and over utilization of surface and ground water – Floods, drought, conflicts over water, dams – benefits and problems - Mineral resources: Use and exploitation, environmental effects of extracting and using mineral resources, case studies. - Food resources: World food problems, changes caused by agriculture and overgrazing, effects of modern agriculture, fertilizer-pesticide problems, water logging, salinity, case studies. - Energy resources: Growing energy needs, renewable and non-renewable energy sources use of alternate energy sources. Case studies. Land resources: Land as a resource, land degradation, man induced landslides, soil erosion and desertification. Role of an individual in conservation of natural resources. Equitable use of resources for sustainable lifestyles.

**UNIT - II**

**ECOSYSTEMS** : Concept of an ecosystem. - Structure and function of an ecosystem. - Producers, consumers and decomposers. - Energy flow in the ecosystem - Ecological succession. - Food chains, food webs and ecological pyramids. - Introduction, types, characteristic features, structure and function of the following ecosystem:

- Forest ecosystem
- Grassland ecosystem
- Desert ecosystem
- Aquatic ecosystems (ponds, streams, lakes, rivers, oceans, estuaries)

**UNIT - III**

**BIODIVERSITY AND ITS CONSERVATION** : Introduction - Definition: genetic, species and ecosystem diversity. - Bio-geographical classification of India - Value of biodiversity: consumptive use, productive use, social, ethical, aesthetic and option values - . Biodiversity at global, National and local levels. - . India as a mega-diversity nation - Hot-spots of biodiversity - Threats to biodiversity: habitat loss, poaching of wildlife, man-wildlife conflicts. - Endangered and endemic species of India – Conservation of biodiversity: In-situ and Ex-situ conservation of biodiversity.

**UNIT - IV**

**ENVIRONMENTAL POLLUTION:** Definition, Cause, effects and control measures of:

- Air pollution
- Water pollution
- Soil pollution
- Marine pollution
- Noise pollution
- Thermal pollution
- Nuclear hazards

**SOLID WASTE MANAGEMENT:** Causes, effects and control measures of urban and industrial wastes. - Role of an individual in prevention of pollution. - Pollution case studies. - Disaster management: floods, earthquake, cyclone and landslides.

**UNIT - V**

**SOCIAL ISSUES AND THE ENVIRONMENT** : From Unsustainable to Sustainable development -Urban problems related to energy -Water

conservation, rain water harvesting, watershed management - Resettlement and rehabilitation of people; its problems and concerns. Case Studies -Environmental ethics: Issues and possible solutions. -Climate change, global warming, acid rain, ozone layer depletion, nuclear accidents and holocaust. Case Studies. - Wasteland reclamation. -Consumerism and waste products. - Environment Protection Act. -Air (Prevention and Control of Pollution) Act. -Water (Prevention and control of Pollution) Act -Wildlife Protection Act -Forest Conservation Act -Issues involved in enforcement of environmental legislation. -Public awareness.

**HUMAN POPULATION AND THE ENVIRONMENT:** Population growth, variation among nations. Population explosion - Family Welfare Programme. -Environment and human health. -Human Rights. -Value Education. -HIV/AIDS. -Women and Child Welfare. -Role of information Technology in Environment and human health. -Case Studies.

**FIELD WORK :** Visit to a local area to document environmental assets River /forest grassland/hill/mountain -Visit to a local polluted site-Urban/Rural/industrial/ Agricultural Study of common plants, insects, birds. -Study of simple ecosystemspond, river, hill slopes, etc.

#### TEXT BOOK:

- 1 Textbook of Environmental Studies for Undergraduate Courses by Erach Bharucha for University Grants Commission.,UniversitiesPress
- 2 Environmental Studies by R. Rajagopalan, Oxford University Press.

#### REFERENCE:

1. Textbook of Environmental Sciences and Technology by M. Anji Reddy, BS Publication.

## JNTUH COLLEGE OF ENGINEERING HYDERABAD

### I Year B.Tech. Chem. Engg. II-Sem

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### COMPUTATIONAL MATHEMATICS (Common to all Branches)

**Pre Requisites:** NIL

#### Objectives:

- This course aims at providing the student with the concepts of matrices, numerical techniques and curve fitting.

#### Outcomes:

At the end of the course, the student will be able to:

- analyze engineering problems using the concepts of Matrices and Numerical Methods.

#### UNIT-I: Matrices and Linear Transformations (8 lectures)

Real matrices – Symmetric, skew – symmetric, orthogonal. Complex matrices: Hermitian, Skew-Hermitian and Unitary Matrices. Idempotent matrix, Finding rank of a matrix by reducing to Echelon and Normal forms. Consistency of system of linear equations (homogeneous and non- homogeneous) using the rank of a matrix.

Cayley-Hamilton Theorem (without Proof) – Verification. Finding inverse of a matrix and powers of a matrix by Cayley-Hamilton theorem, Linear dependence and Independence of Vectors. Linear Transformation – Orthogonal Transformation. Eigen values and Eigen vectors of a matrix. Properties of Eigen values and Eigen vectors of matrices. Diagonalization of matrix – Quadratic forms upto three variables- Reduction of quadratic form to canonical form, Rank – Positive definite, negative definite – semi definite – index – signature of quadratic form.

#### UNIT-II: Interpolation and Curve fitting (5 lectures)

**Interpolation:** Introduction- Errors in Polynomial Interpolation – Finite differences- Forward Differences- Backward differences –Central differences – Symbolic relations and separation of symbols- Difference Equations – Differences of a polynomial-Newton's formulae for interpolation –Interpolation with unevenly spaced points-Lagrange's Interpolation formula. **Curve fitting:** Fitting a straight line –Second degree curve-exponential curve-power curve by method of least squares.

**UNIT–III: Numerical techniques (5 lectures)**

Solution of Algebraic and Transcendental Equations and Linear system of equations.

Introduction – Graphical interpretation of solution of equations .The Bisection Method – The Method of False Position – The Iteration Method – Newton-Raphson Method .

Solving system of non-homogeneous equations by L-U Decomposition method(Crout's Method)Jacobi's and Gauss-Seidel Iteration method

**UNIT- IV: Numerical Differentiation, Integration: (5 lectures)**

Numerical differentiation, Numerical integration – Trapezoidal rule, Simpson's  $1/3^{rd}$  and  $3/8$  Rule , Generalized Quadrature.

**UNIT – V: Numerical solutions of First order differential equations (5 lectures)**

Numerical solution of Ordinary Differential equations: Solution by Taylor's series method –Picard's Method of successive Approximation- single step methods-Euler's Method-Euler's modified method, Runge-Kutta Methods.

**Text Books:**

- 1) INTRODUCTORY METHODS OF NUMERICAL ANALYSIS BY SS SASTRY
- 2) NUMERICAL AND STATISTICAL METHODS WITH PROGRAMMING IN C BY SUJATHA SINHA AND SUBHABRATA DINDA, SCITEC PUBLISHERS.
- 3) NUMERICAL METHODS, PRINCIPLES, ANALYSIS AND ALGORITHMS BY SRIMANTAPAL & SUBODH C. BHUNIA, OXFORD UNIVERSITY PRESS.

**References:**

- 1) ADVANCED ENGINEERING MATHEMATICS BY ALAN JEFFERY
- 2) APPLIED NUMERICAL METHODS USING MATLAB BY RAO.V.DUKKIPATI, NEW AGE PUBLISHERS
- 3) NUMERICAL METHODS IN SCIENCE AND ENGINEERING –APRACTICAL APPROACH BY S.RAJASEKHARAN, S.CHAND PUBLICATIONS

**JNTUH COLLEGE OF ENGINEERING HYDERABAD****I Year B.Tech. Chem. Engg. II-Sem**

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**ENGINEERING WORKSHOP**

Pre-requisites: **Practical skill**

**Objectives:**

- To Study of different hand operated power tools, uses and their demonstration.
- To gain a good basic working knowledge required for the production of various engineering products.
- To provide hands on experience about use of different engineering materials, tools, equipments and processes those are common in the engineering field.
- To develop a right attitude, team working, precision and safety at work place.
- It explains the construction, function, use and application of different working tools, equipment and machines.
- To study commonly used carpentry joints.
- To have practical exposure to various welding and joining processes.
- Identify and use marking out tools, hand tools, measuring equipment and to work to prescribed tolerances.
- To understanding the computer hardware and practice the Assembly of computer parts.
- To practice the process of Installation of operating system windows.

**Outcomes:**

At the end of the course, the student will be able to:

- Better understanding the process of assembly of computer parts and installation of different software's.
- Study and practice on machine tools and their operations
- Practice on manufacturing of components using workshop trades including plumbing, fitting, carpentry, foundry, house wiring and welding.
- Identify and apply suitable tools for different trades of Engineering processes including drilling, material removing, measuring, chiseling.
- Apply basic electrical engineering knowledge for house wiring practice.



## I. TRADES FOR EXERCISES :

(Any **six** trades from the following with minimum of **two** exercises in each trade)

1. Carpentry
2. Fitting
3. Tin-Smithy
4. Black Smithy
5. House-wiring
6. Foundry
7. Plumbing

## II. Trades for Demonstration &amp; Exposure

1. Demonstration of power tools & wiring
2. Welding
3. Machine Shop

III. **IT Workshop I:** Computer hardware, identification of parts, Disassembly, Assembly of computer to working condition, simple diagnostic exercises.

**IT Workshop II:** Installation of operating system windows and linux simple diagnostic exercises.

## JNTUH COLLEGE OF ENGINEERING HYDERABAD

I Year B.Tech. Chem. Engg. II-Sem

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## ENGLISH LANGUAGE COMMUNICATION SKILLS LAB

The **Language Lab** focuses on the production and practice of sounds of language and familiarises the students with the use of English in everyday situations and contexts.

**Objectives**

- ✎ To facilitate computer-aided multi-media instruction enabling individualized and independent language learning
- ✎ To sensitise the students to the nuances of English speech sounds, word accent, intonation and rhythm
- ✎ To bring about a consistent accent and intelligibility in their pronunciation of English by providing an opportunity for practice in speaking
- ✎ To improve the fluency in spoken English and neutralize mother tongue influence
- ✎ To train students to use language appropriately for interviews, group discussion and public speaking

**Learning Outcomes**

- 👉 Better Understanding of nuances of language through audio- visual experience and group activities
- 👉 Neutralization of accent for intelligibility
- 👉 Speaking with clarity and confidence thereby enhancing employability skills of the students

## SYLLABUS

English Language Communication Skills Lab shall have two parts:

- a. Computer Assisted Language Learning (CALL) Lab
- b. Interactive Communication Skills (ICS) Lab

The following course content is prescribed for the **English Language Communication Skills Lab**

**Exercise – I**

**CALL Lab:** Introduction to Phonetics – Speech Sounds – Vowels and Consonants

**ICS Lab:** Ice-Breaking activity and JAM session

Articles, Prepositions, Word formation- Prefixes & Suffixes, Synonyms & Antonyms

### Exercise – II

**CALL Lab:** Structure of Syllables - Past Tense Marker and Plural Marker – Weak Forms and Strong Forms - Consonant Clusters.

**ICS Lab:** Situational Dialogues – Role-Play- Expressions in Various Situations – Self-introduction and Introducing Others – Greetings – Apologies – Requests – Social and Professional Etiquette - Telephone Etiquette.

Concord (Subject in agreement with verb) and Words often misspelt-confused/misused

### Exercise - III

**CALL Lab:** Minimal Pairs- Word accent and Stress Shifts- Listening Comprehension.

**ICS Lab:** Descriptions- Narrations- Giving Directions and guidelines. Sequence of Tenses, Question Tags and One word substitutes.

### Exercise – IV

**CALL Lab:** Intonation and Common errors in Pronunciation.

**ICS Lab:** Extempore- Public Speaking

Active and Passive Voice, –Common Errors in English, Idioms and Phrases

### Exercise – V

**CALL Lab:** Neutralization of Mother Tongue Influence and Conversation Practice

**ICS Lab:** Information Transfer- Oral Presentation Skills

Reading Comprehension and Job Application with Resume preparation.

### Minimum Requirement of infrastructural facilities for ELCS Lab:

#### 1. Computer Assisted Language Learning (CALL) Lab:

**The Computer aided Language Lab** for 40 students with 40 systems, one master console, LAN facility and English language software for self- study by learners.

#### **System Requirement (Hardware component):**

*Computer network with Lan with minimum 60 multimedia systems with the following specifications:*

- i) P – IV Processor
  - a) Speed – 2.8 GHZ
  - b) RAM – 512 MB Minimum
  - c) Hard Disk – 80 GB
- ii) Headphones of High quality

#### 2. Interactive Communication Skills (ICS) Lab :

**The Interactive Communication Skills Lab:** A Spacious room with movable chairs and audio-visual aids with a Public Address System, a T. V., a digital stereo –audio & video system and camcorder etc.

#### Suggested Software:

- ❖ **Cambridge Advanced Learners' English Dictionary with CD.**
- ❖ **Grammar Made Easy by Darling Kindersley**
- ❖ **Punctuation Made Easy by Darling Kindersley**
- ❖ Clarity Pronunciation Power – Part I
- ❖ Clarity Pronunciation Power – part II
- ❖ **Oxford Advanced Learner's Compass, 8<sup>th</sup> Edition**
- ❖ **DELTA's key to the Next Generation TOEFL Test: Advanced Skill Practice.**
- ❖ Lingua TOEFL CBT Insider, by Dreamtech
- ❖ TOEFL & GRE (KAPLAN, AARCO & BARRONS, USA, Cracking GRE by CLIFFS)
- ❖ **English in Mind (Series 1-4), Herbert Puchta and Jeff Stranks with Meredith Levy, Cambridge**
- ❖ **English Pronunciation in Use** (Elementary, Intermediate, Advanced) Cambridge University Press
- ❖ Raman, M & Sharma, S. 2011. Technical Communication, OUP
- ❖ Sanjay Kumar & Pushp Lata. 2011. Communication Skills, OUP

#### SUGGESTED READING:

1. Rama Krishna Rao, A. *et al. English Language Communication Skills – A Reader cum Lab Manual Course Content and Practice.* Chennai: Anuradha Publishers
2. Suresh Kumar, E. & Sreehari, P. 2009. *A Handbook for English Language Laboratories.* New Delhi: Foundation
3. *Speaking English Effectively* 2<sup>nd</sup> Edition by Krishna Mohan and N. P. Singh, 2011. Macmillan Publishers India Ltd. Delhi.
4. Sasi Kumar, V & Dhamija, P.V. *How to Prepare for Group Discussion and Interviews.* Tata McGraw Hill
5. Spoken English: A Manual of Speech and Phonetics by R. K. Bansal & J. B. Harrison. 2013. Orient Blackswan. Hyderabad.
6. *English Pronunciation in Use. (Elementary, Intermediate & Advance).* Cambridge: CUP
7. Chris Redston, Gillie Cunningham, Jan Bell. *Face to Face* (2<sup>nd</sup> Edition). Cambridge University Press
8. Nambiar, K.C. 2011. *Speaking Accurately. A Course in International Communication.* New Delhi : Foundation

9. Soundararaj, Francis. 2012. *Basics of Communication in English*. New Delhi: Macmillan
10. A textbook of English Phonetics for Indian Students by T. Balasubramanian (Macmillan)

### DISTRIBUTION AND WEIGHTAGE OF MARKS

#### **English Language Laboratory Practical Examination:**

1. The practical examinations for the English Language Laboratory shall be conducted as per the University norms prescribed for the core engineering practical sessions.
2. For the Language lab sessions, there shall be a continuous evaluation during the year for 30 sessional marks and 70 semester-end Examination marks. Of the 30 marks, 20 marks shall be awarded for day-to-day work and 10 marks to be awarded by conducting Internal Lab Test(s). The year- end Examination shall be conducted by the teacher concerned with the help of another member of the staff of the same department of the same institution.

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### JNTUH COLLEGE OF ENGINEERING HYDERABAD

I Year B.Tech. Chem. Engg. I-Sem

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### COMPUTATIONAL MATHEMATICS LAB (Common to all Branches)

#### **UNIT- I: Interpolation**

##### **Programming Tasks:**

- A) Write a program to determine y for a given x, if two arrays of x and y of same size are given (using Newton's interpolation both forward and backward)
- B) Write a program to determine y for a given x, if two arrays of x and y of same size are given.(using Lagrange 's interpolation)
- C) Write a program to determine y for a given x, if two arrays of x and y of same size are given.(using Gauss interpolation)  
(Selection criteria of the interpolation formula are important.)

#### **UNIT- 1I: Curve fitting**

##### **Programming Tasks:**

- A) Write a program to find a line of best fit from the given two arrays of x and y of same size.
- B) Write a program to find a curve of the form  $y = Ae^{Bx}$  from the given two arrays of x and y of same size.
- C) Write a program to find a curve of the form  $y = Ax^B$  from the given two arrays of x and y of same size.
- D) Write a program to find a curve of the form  $y = Ax^2 + Bx + C$  from the given two arrays of x and y of same size.

#### **UNIT- 1II: Solution of Algebraic and Transcendental Equations**

##### **Programming Tasks:**

- A) Write a program to find the root of a given equation using bisection method.  
(Write this program such that the initial values given to the system are not usable, then the system should ask us to give new set of initial values).
- B) Write a program to find the root of a given equation using method of false position(regula false position).
- C) Write a program to find the root of a given equation using iteration method.
- D) Write a program to find the root of a given equation using Newton Rophson method.

#### **UNIT- IV: Linear system of equations**

##### **Programming Tasks:**

- A) Write a program to find the solution of given system of linear equations using L- U decomposition method.
- B) Write a program to find the solution of given system of linear equations using jacobi's method.
- C) Write a program to find the solution of given system of equations using Gauss sidel iteration method.
- D) Write a program to find the solution of given system of equations using Gauss Jordan elimination method.

#### **UNIT-V:Numerical Differentiation, Integration and Numerical solutions of First order differential equations**

##### **Programming Tasks:**

- A) Write a program to evaluate definite integral using trapezoidal rule, Simpson's  $1/3^{\text{rd}}$  rule and  $3/8^{\text{th}}$  rule.
- B) Write a program to solve a given differential equation using Taylor's series.
- C) Write a program to solve a given differential equation Euler's and modified Eulers method.
- D) Write a program to solve a given differential equation using Ruge-Kutta method.